

# BEHAVIOUR, PROPERTIES AND STRUCTURES OF COMPLEX FLUIDS UPON SPREADING

CHRISTOPHE KUSINA

SUPERVISED BY :

ANNIE COLIN (CBI-MIE ESPCI PARIS)

ODILE AUBRUN, JEAN-BAPTISTE BOITTE, DANIELLE LE VERGE (L'ORÉAL)

# Industrial context

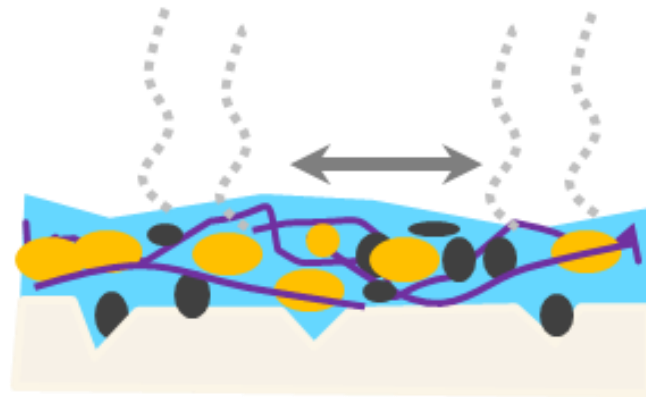
*Life cycle of cosmetic products*

## Packaging, storing



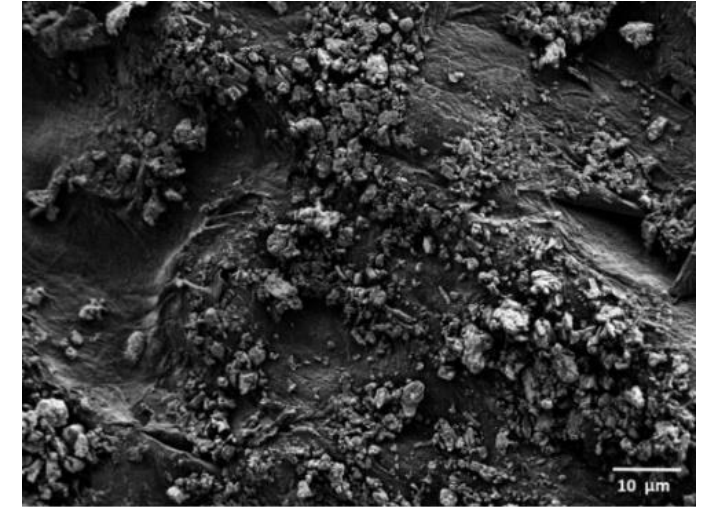
Bulk **stability** of formulations  
Stability over time

## Application



**Spreading dynamic** and behaviour  
of complex fluids under **stress** and  
during drying

## Efficiency, performance



Deposit **characterization**  
**Distribution** of particles

# Industrial context

*Life cycle of cosmetic products*

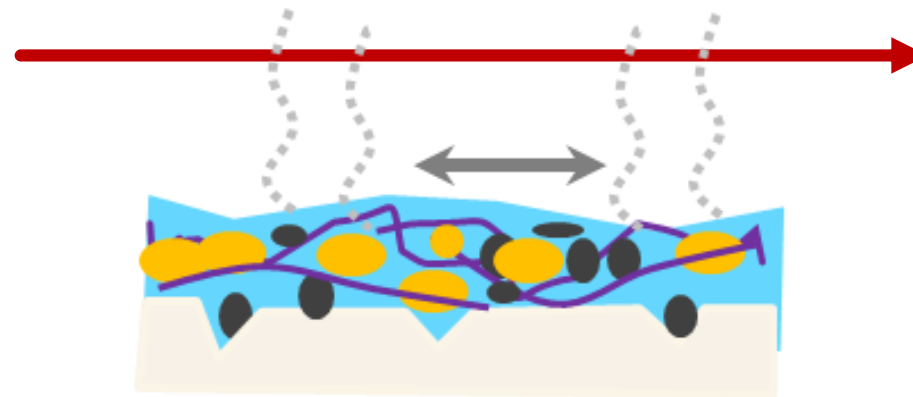
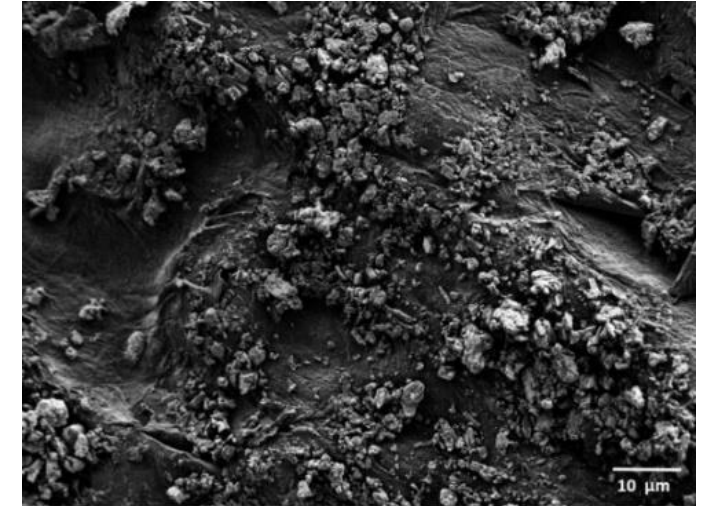
Packaging, storing



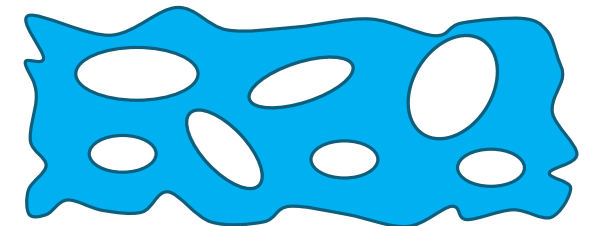
Application



Efficiency, performance



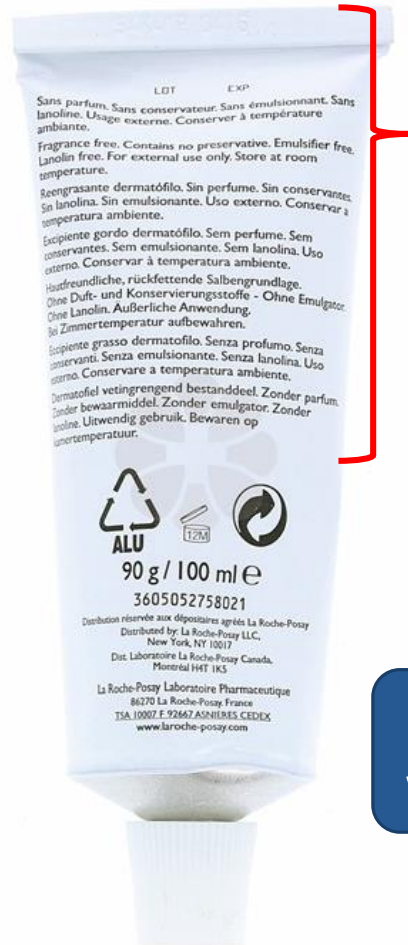
Homogeneous coating



Heterogeneous coating

What can we do to get a **homogeneous coating of cosmetics on skin** ?

# What is a cosmetic formulation made of ?



Volatil phase (aqueous)



Water

Active non-volatil phase



Glycerol, oils, waxes  
Pigments, UV filters

Texture agents



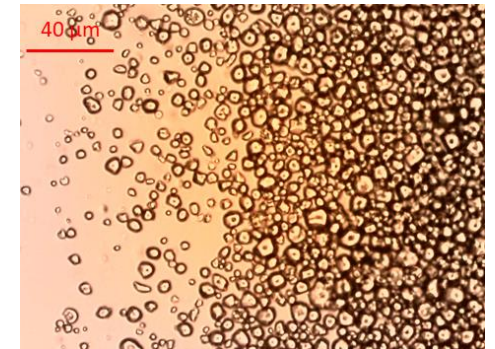
Carbopol  
Particles (starch, silica)

Cosmetics are **complex fluids**  
with **non-newtonian behaviours**

Hundreds of ingredients



Yield stress fluids



Shear-thickening fluids

# Introduction to non-Newtonian behaviours

Newtonian fluids (simple fluids)

Monophasic

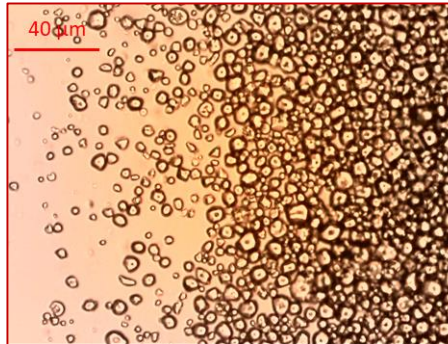
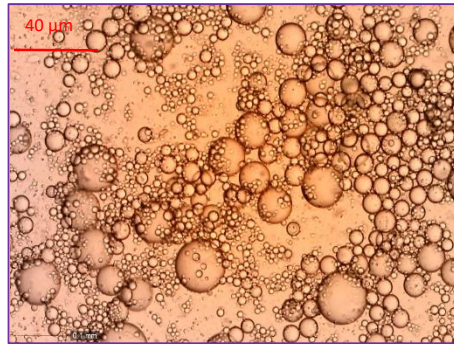


$$\eta(\dot{\gamma}) = cst$$

Their **viscosity is constant** whatever the shear they undergo

Complex fluids

Multiple phases



$$\eta(\dot{\gamma}) \neq cst$$

Their **viscosity varies** with the shear they undergo



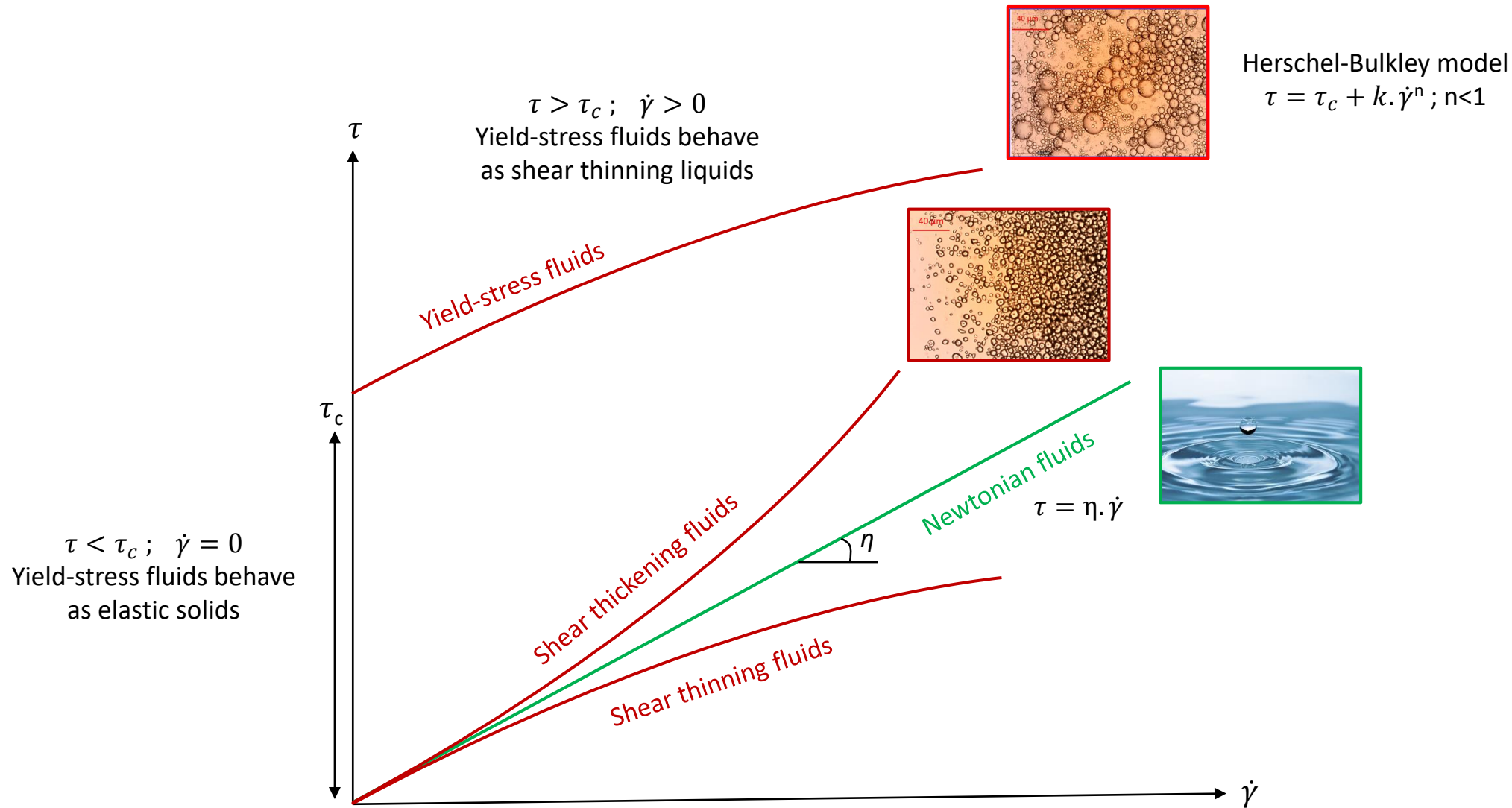
Non-Newtonian behaviour

Structure-Flow coupling

Yield stress fluids

Shear-thickening fluids

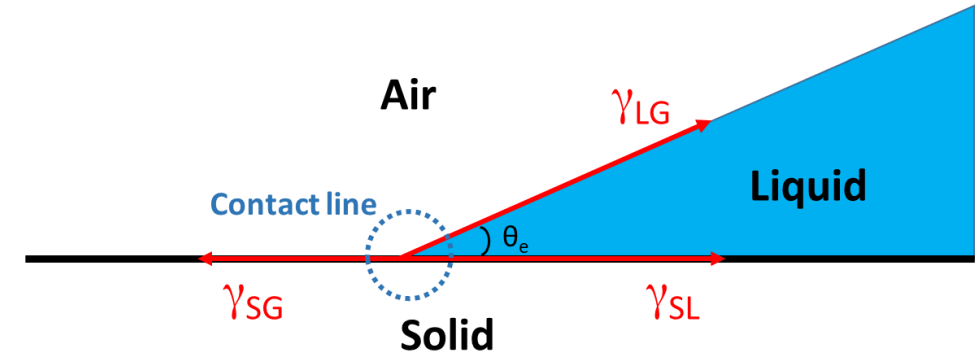
# Introduction to non-Newtonian behaviours



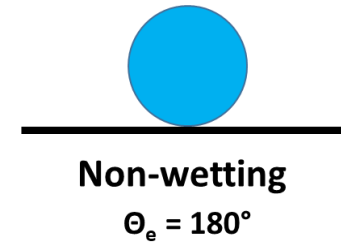
Spreading of non-Newtonian fluids is new: interesting for academic research

Everyday problem

# Spontaneous spreading of fluids: wetting properties



Young's law  $\gamma_{SG} = \gamma_{SL} + \gamma_{LG} \cos \theta_e$



Non-wetting

$$\theta_e = 180^\circ$$



Partial wetting

$$0^\circ < \theta_e < 180^\circ$$



Total wetting

$$\theta_e = 0^\circ$$

Spreading coefficient

$$S = \gamma_{SG} - \gamma_{LG} - \gamma_{SL}$$

**Spontaneous spreading of Newtonian fluids well understood**

- Young, T. An Essay on the Cohesion of Fluids. *Philos. Trans. R. Soc. London* **95**, 65–87 (1805).  
Harkins, W. D. & Feldman, A. Films. The Spreading Of Liquids And the Spreading Coefficient. *Am. Chem. Soc.* **44**, (1922).  
Zisman, W. A. Relation of the Equilibrium Contact Angle to Liquid and Solid Constitution. *Adv. Chem. Ser.* (1964).  
Bergeron, V., Fagan, M. E. & Radke, C. J. Generalized entering coefficients: a criterion for foam stability against oil in porous media. *Langmuir* **9**, 1704–1713 (1993).

# Forcing the spreading of a non-wetting fluid



# Forcing the spreading of a non-wetting fluid

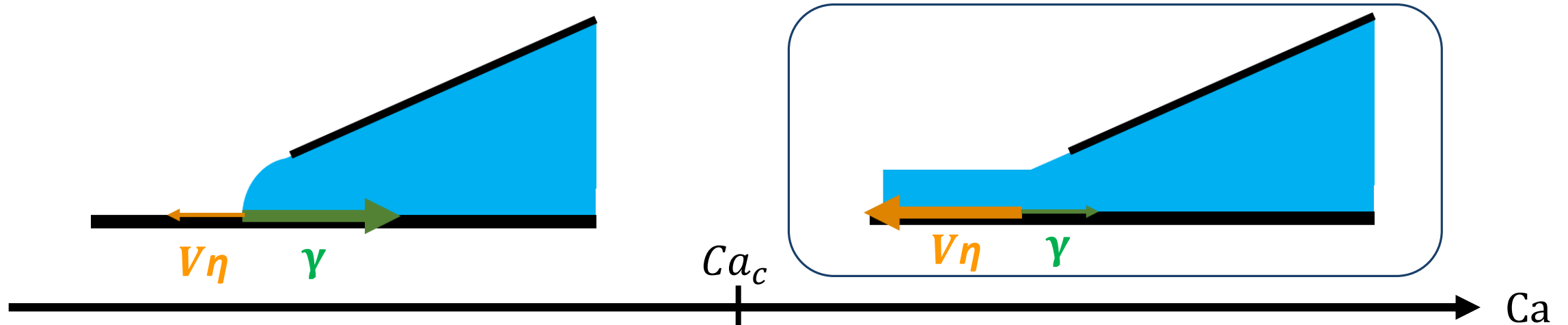
$$Ca = \frac{V\eta}{\gamma}$$

$U$  : Spreading velocity  
 $\eta$  : Viscosity of the fluid  
 $\gamma$  : Liquid/air surface tension

$$Ca_c = \frac{\theta_e^3}{9 \ln \frac{l_c}{l'}}$$

$\theta_e$  : Equilibrium contact angle  
 $l_c$  : Capillary length  
 $l'$  : Molecular length

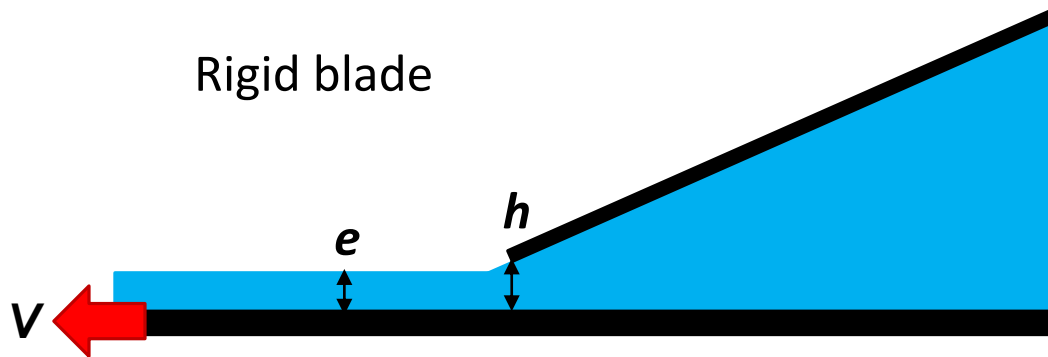
Comparison between  
**viscous stress** and **surface tension**



What fixes the thickness of the formed film ?

# Spreading of Newtonian fluids

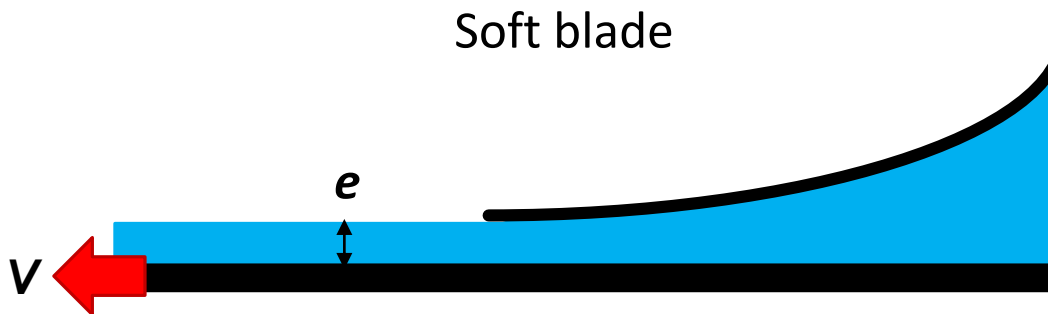
## Blade-Coating using rigid and soft blades



No slip and small angle  
Only depends on the geometry

$$e \sim h/2$$

Greener, Y. & Middleman, S. Blade-coating of a viscoelastic fluid. *Polym. Eng. Sci.* **14**, 791–796 (1974).



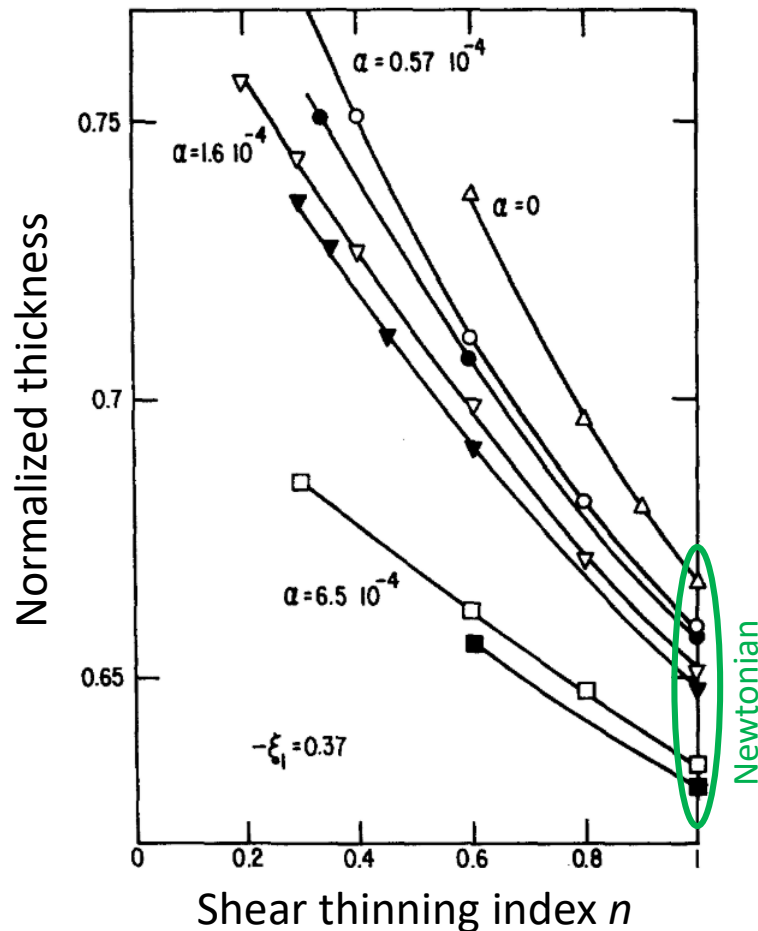
Coupling flow-geometry

$$e \sim L \left( \frac{\eta V L^2}{B} \right)^{3/4}$$

$V$  : Spreading velocity  
 $\eta$  : Viscosity of the fluid  
 $L$  : Length of the blade  
 $B$  : Rigidity of the blade

# Spreading of non-Newtonian fluids

Blade-Coating using rigid blades



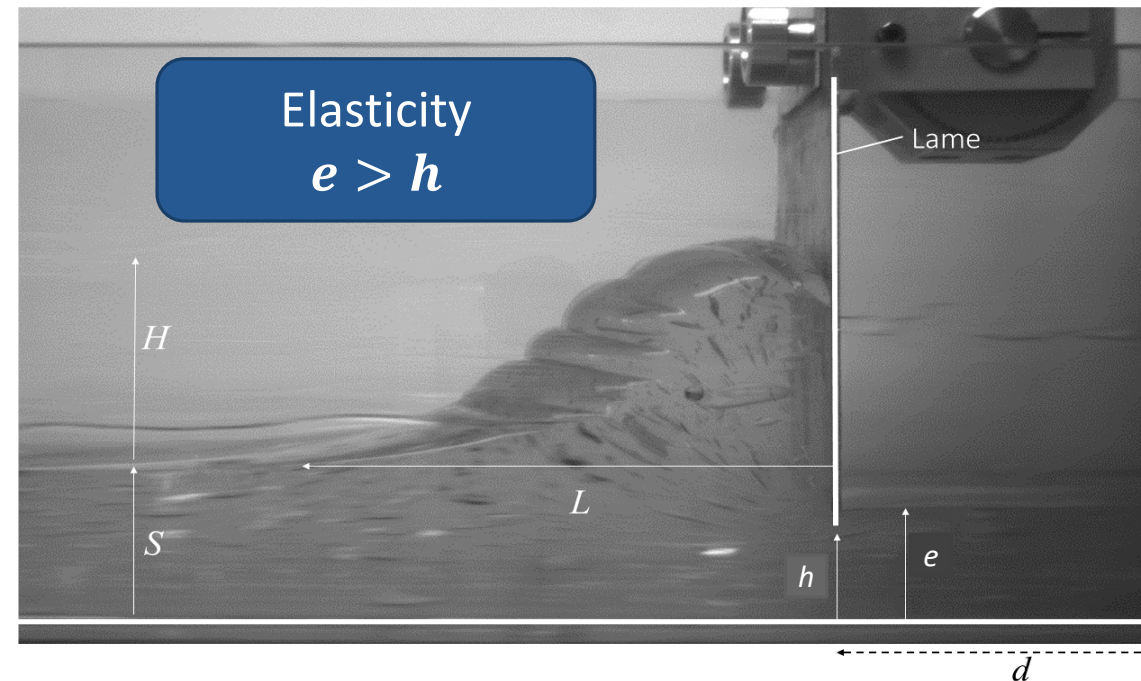
Films are **thicker** than those made of newtonian fluids

$$e > h/2$$

$e$  decreases for increasing  $V$

$e$  increases with shear thinning behaviour

Carbopol gel



Maillard, M. Étalements de fluides à seuil. (2016).

Middleman, S., Greener, J. & Malone, M. *Fundamentals of polymer processing*. (1977).

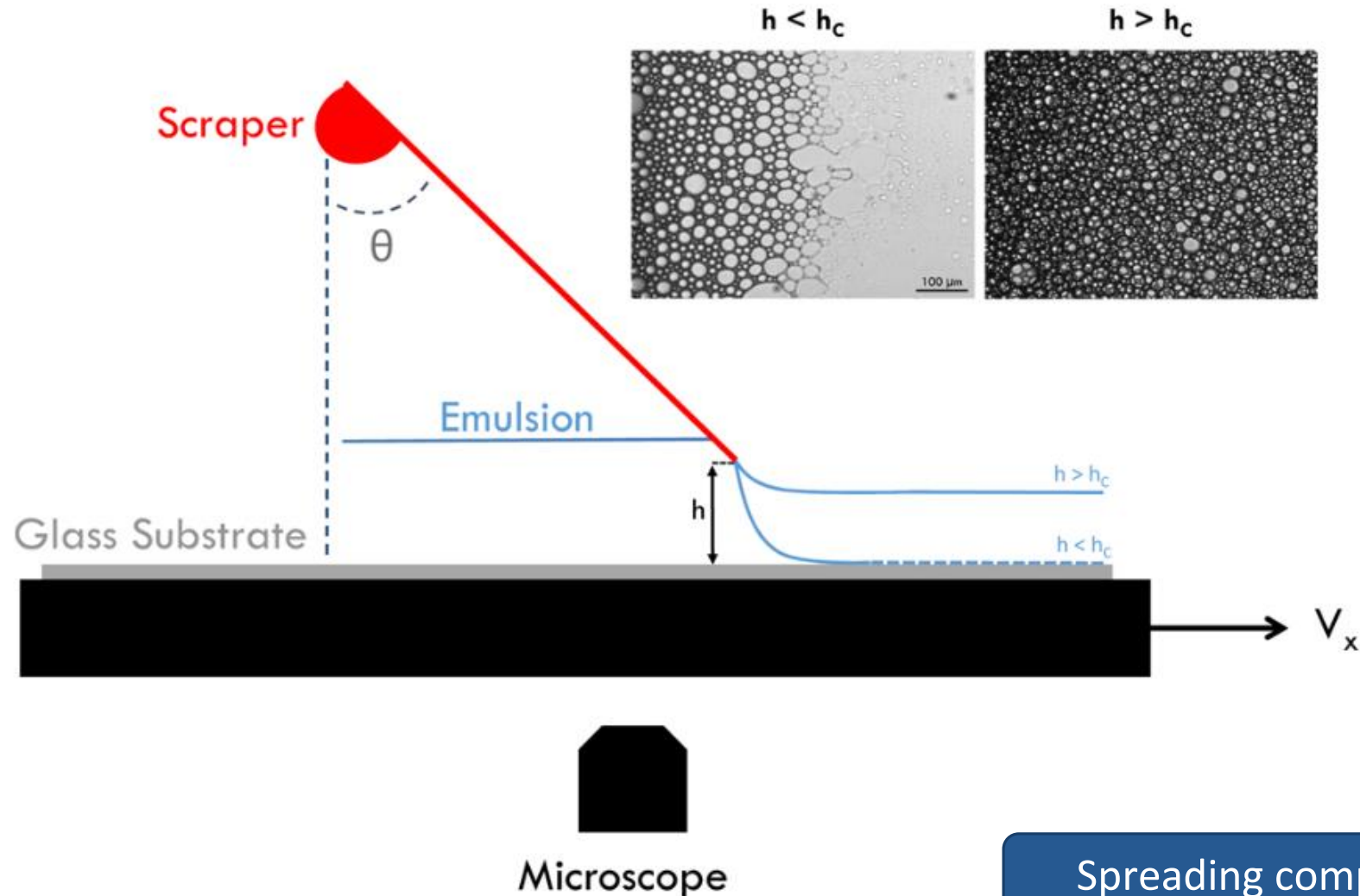
Sullivan, T., Middleman, S. & Keunings, R. *AIChE J.* **33**, 2047–2056 (1987).

Sullivan, T. M. & Middleman, S. *J. Nonnewton. Fluid Mech.* **21**, 13–38 (1986).

Hsu, T. C., Malone, M., Laurence, R. L. & Middleman, S. *Journal Non-Newtonian Fluid Mech.* **18**, 273–294 (1985).

# Spreading of non-Newtonian fluids

*Blade-Coating using rigid blades*

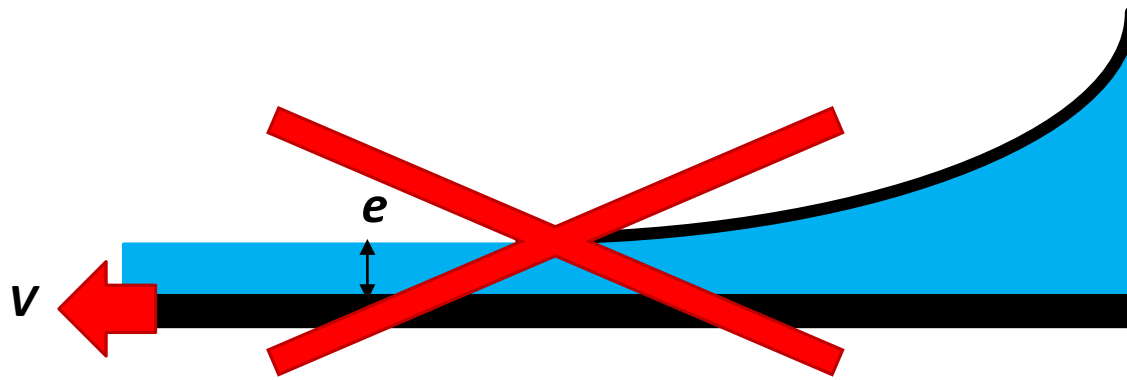


Spreading complex fluid is **not** harmless

Deblais, A., Harich, R., Bonn, D., Colin, A. & Kellay, H. *Langmuir* **31**, 5971–5981 (2015).

# Spreading of non-Newtonian fluids

*Blade-Coating using soft blades*



**No study to predict the thickness of non-Newtonian fluids**

**Clear academical interest**

**And industrial**

# Main questions and strategy

How are films formed using an elastic blade?

- Setting a precise blade-coating experimental set-up
- Observing films using microscopy and profilometry

How can we predict film thickness for complex fluids?

- Carrying coating experiments using different parameters
  - *Spreading velocity*
  - *Rheology of the fluid*
  - *Nature of the substrate*
- Describing the flow under the blade with Stokes' equations

**Part 1**

What do I get if I spread very complex fluids?

- Preparation of a simple cosmetic models
- Study the impact of non-volatile phases and the presence of solid particles on rheological behaviour and spreading
- Setting a more realistic spreading experimental set-up

**Part 2**

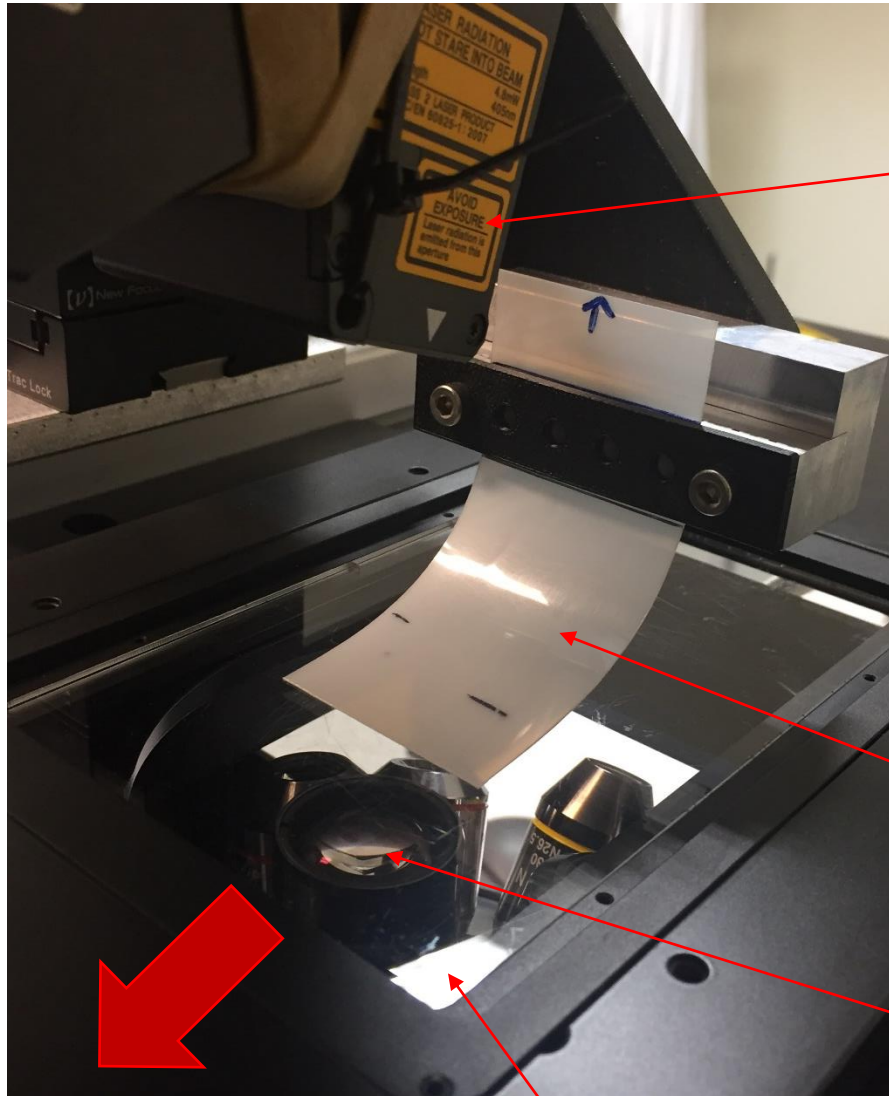


**Part 1:**

***Blade-coating: Spreading a yield-stress fluid***

# Blade Coating under microscopy

## Experimental set-up



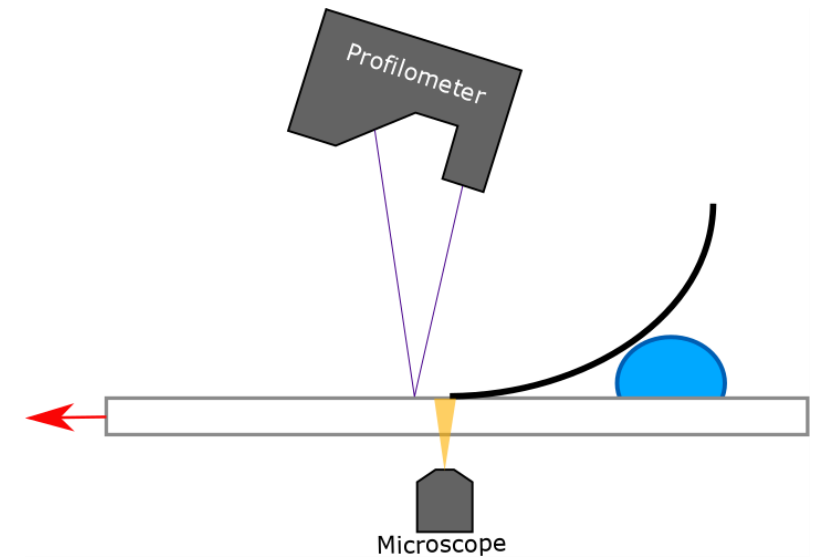
Motion of the stage

PMMA substrate

Profilometer

**Static** Soft Mylar blade  
Length = 5.7 cm; thickness = 125  $\mu\text{m}$ ;  
Width = 4 cm

Microscope

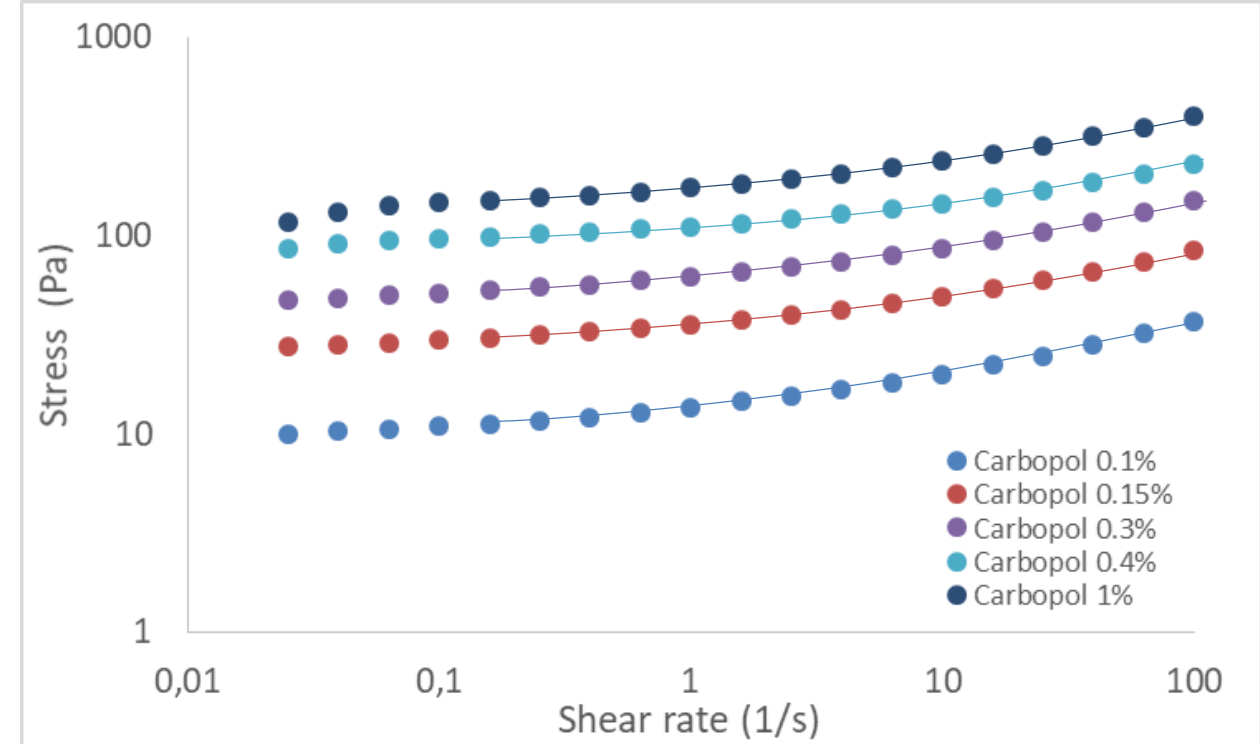


**Easily tunable** : Force, surface tension, ...  
Allows to **observe the film formation** and to **measure the film thickness**

# Yield stress fluids used



Carbopol gels  
= model system of yield stress fluid  
= used often in cosmetic applications



*Flow sweep*

Fitted by a Herschel-Bulkley model:  $\tau = \tau_c + k\dot{\gamma}^n$

# Blade Coating of carbopol gels

Spreading of carbopol gels on a smooth PMMA plate

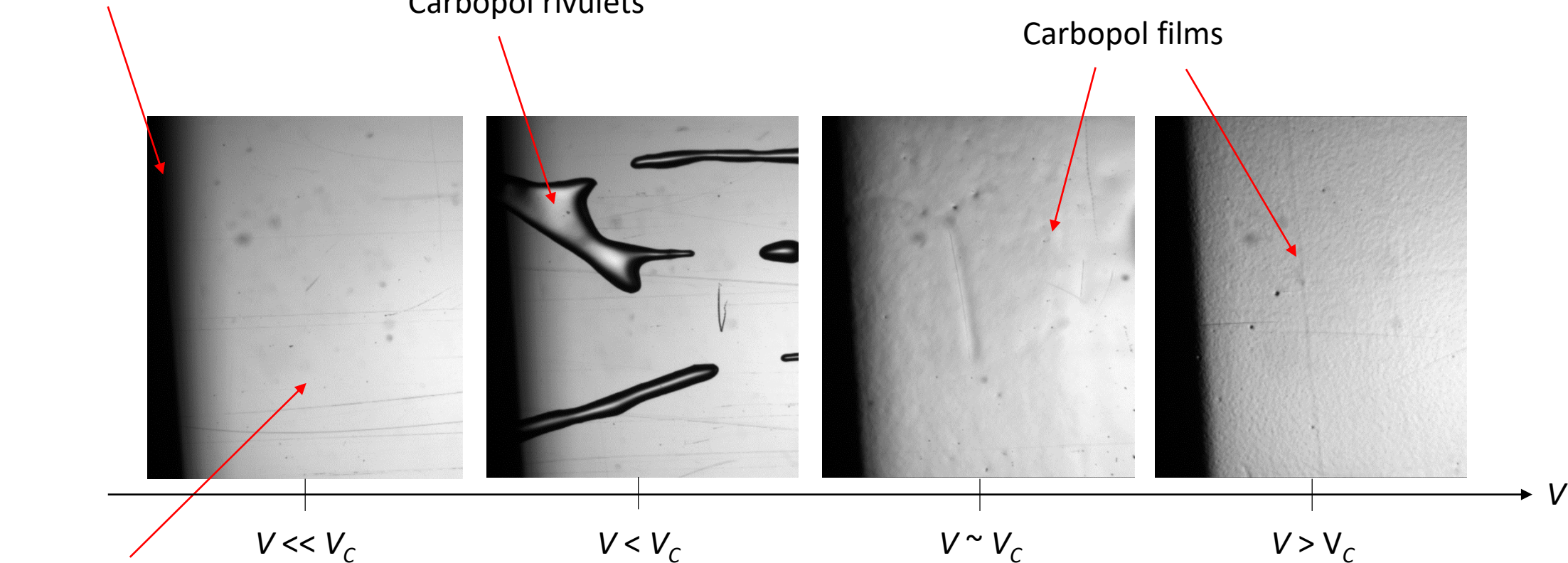
Non-wetting system in slippery conditions

Carbopol gel ( $\tau_c = 46 \text{ Pa}$ )

Scraping blade

Carbopol rivulets

Carbopol films



Substrate

$V \ll V_c$

$V < V_c$

$V \sim V_c$

$V > V_c$

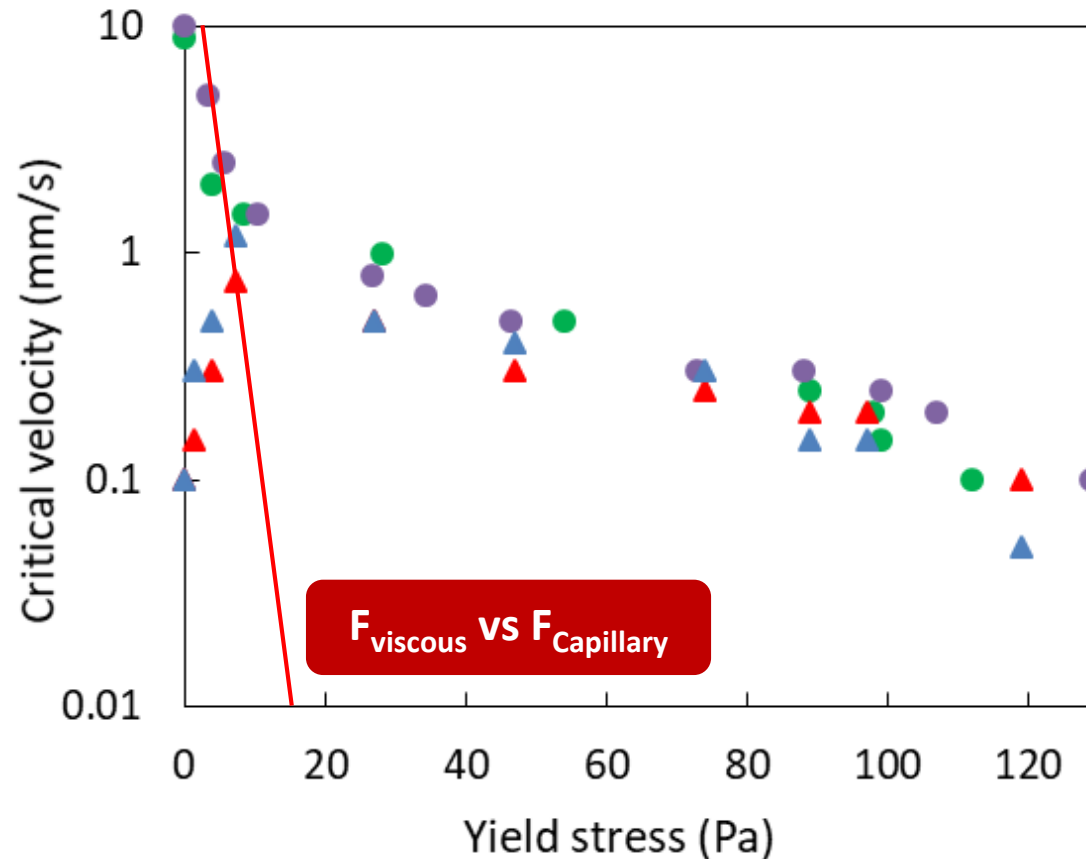
$V$

Spreading occurs only if  $V > V_c$

What does fix  $V_c$  ?

# Blade Coating of carbopol gels

Spreading of carbopol gels on a smooth PMMA plate



- Carbopol (high blade rigidity)
- Carbopol (low blade rigidity)
- ▲ Carbopol + surfactant (low blade rigidity)
- ▲ Carbopol + surfactant (high blade rigidity)

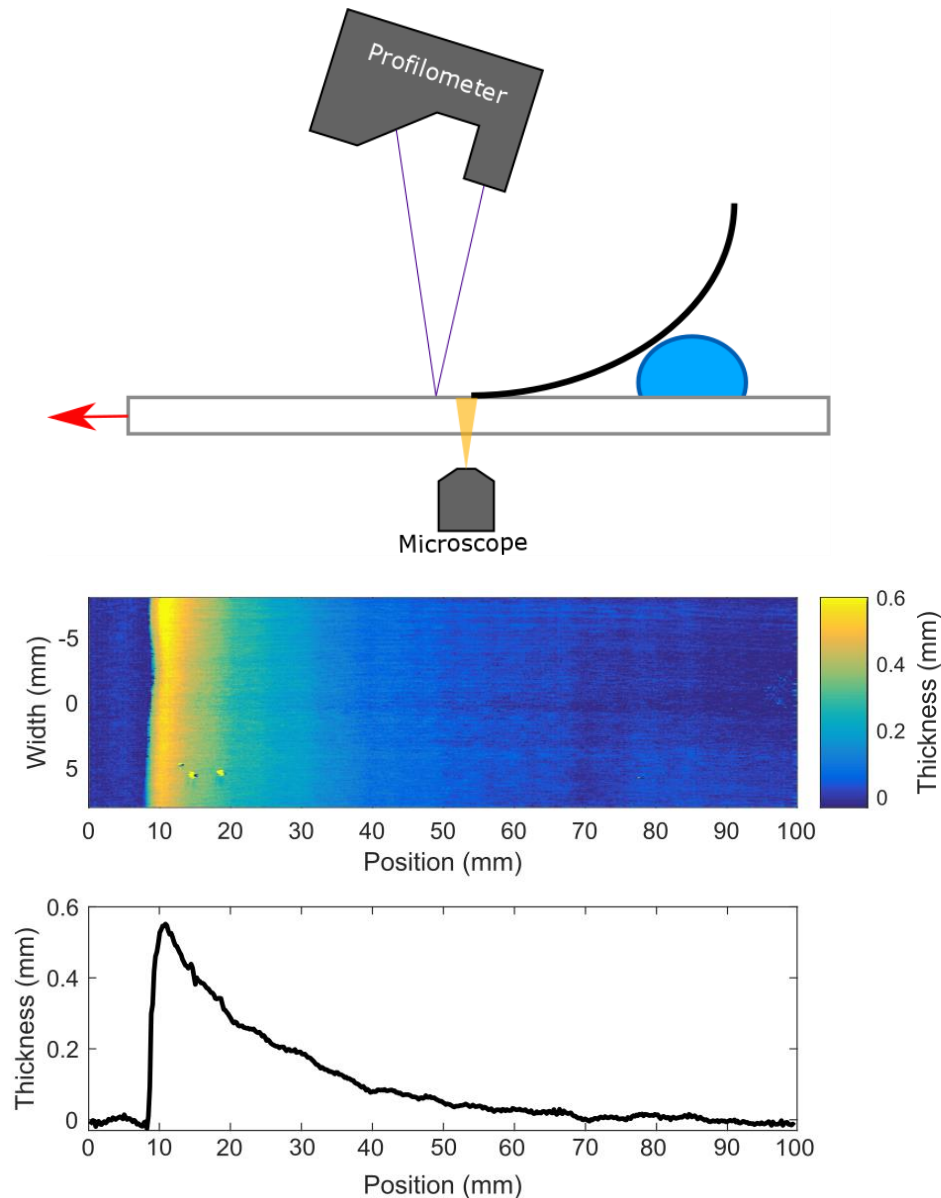
- Yield stress decreases  $V_c$
- Rigidity of the blade : no impact
- Surface tension : no impact for  $\tau_c > 15$  Pa

$V_c$  linked to slippery effect  
Different from what we described before

Rough plates = simpler systems  
Easier to observe the science of spreading

# Blade Coating of carbopol gels

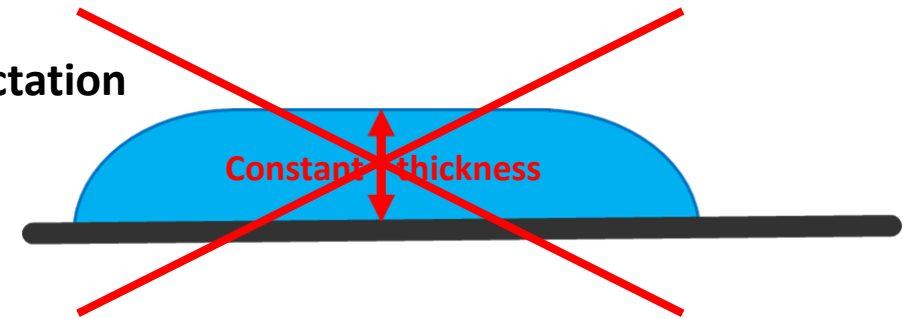
Spreading on a plasma treated PMMA rough plate



## Wetting system in no slip conditions

Film forms whatever is the spreading velocity

Expectation



Reality



*Due to the **finite mass** of the coated **sample***

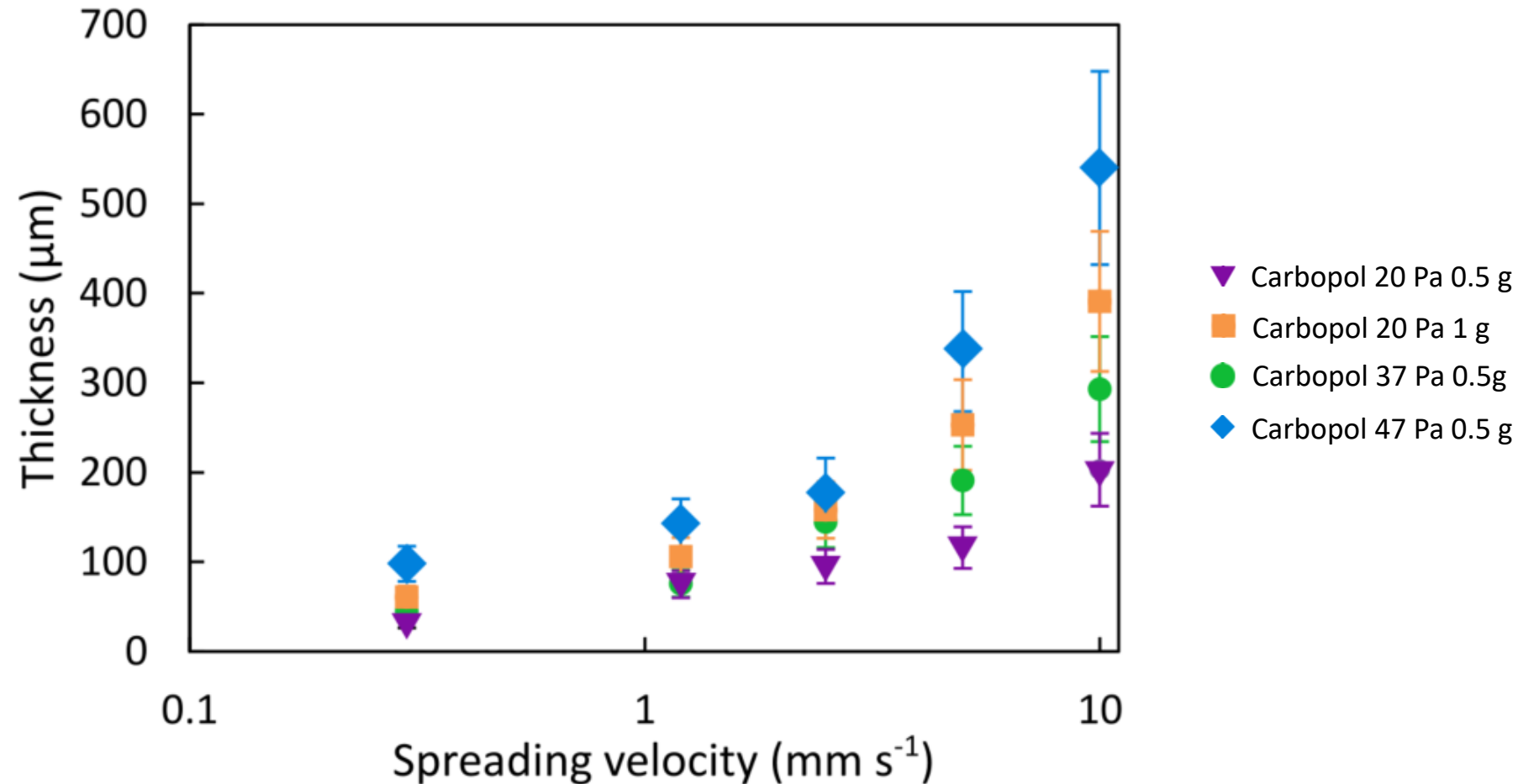
What are the **main parameters** that **fix the thickness** of the film ?

# Blade Coating of carbopol gels

## *Prediction of the film thickness*

Spreading of carbopol gels on a plasma treated PMMA rough plate

Wetting system in no slippery conditions



$e$  increases with  $V$ ,  $\tau_c$  and the mass of sample

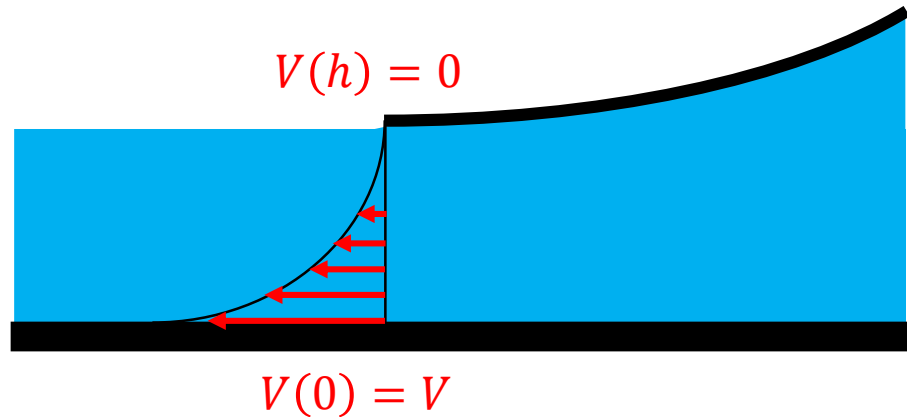
# Blade Coating of carbopol gels

## Prediction of the film thickness

Spreading of carbopol gels on a plasma treated PMMA rough plate

Initial condition :

Raising of the blade in no slip conditions



Stokes' equation under the blade

$$\tau(h) = \frac{\partial P}{\partial x} h - \tau(0)$$

Viscous forces of the fluid = elasticity of the blade

Seiwert (2013) :

$$\frac{\partial P}{\partial x} = \frac{B}{h^{2/3} R_c^{10/3}}$$

$$\tau(h) \approx -\tau_c$$

$$\tau(0) = -\tau_c - k\dot{\gamma}^n$$

$$h \propto \left( \frac{kV^n R_c^{10/3}}{B} \right)^{1/(n+1/3)}$$

$R_c$  : Curvature radius of the blade

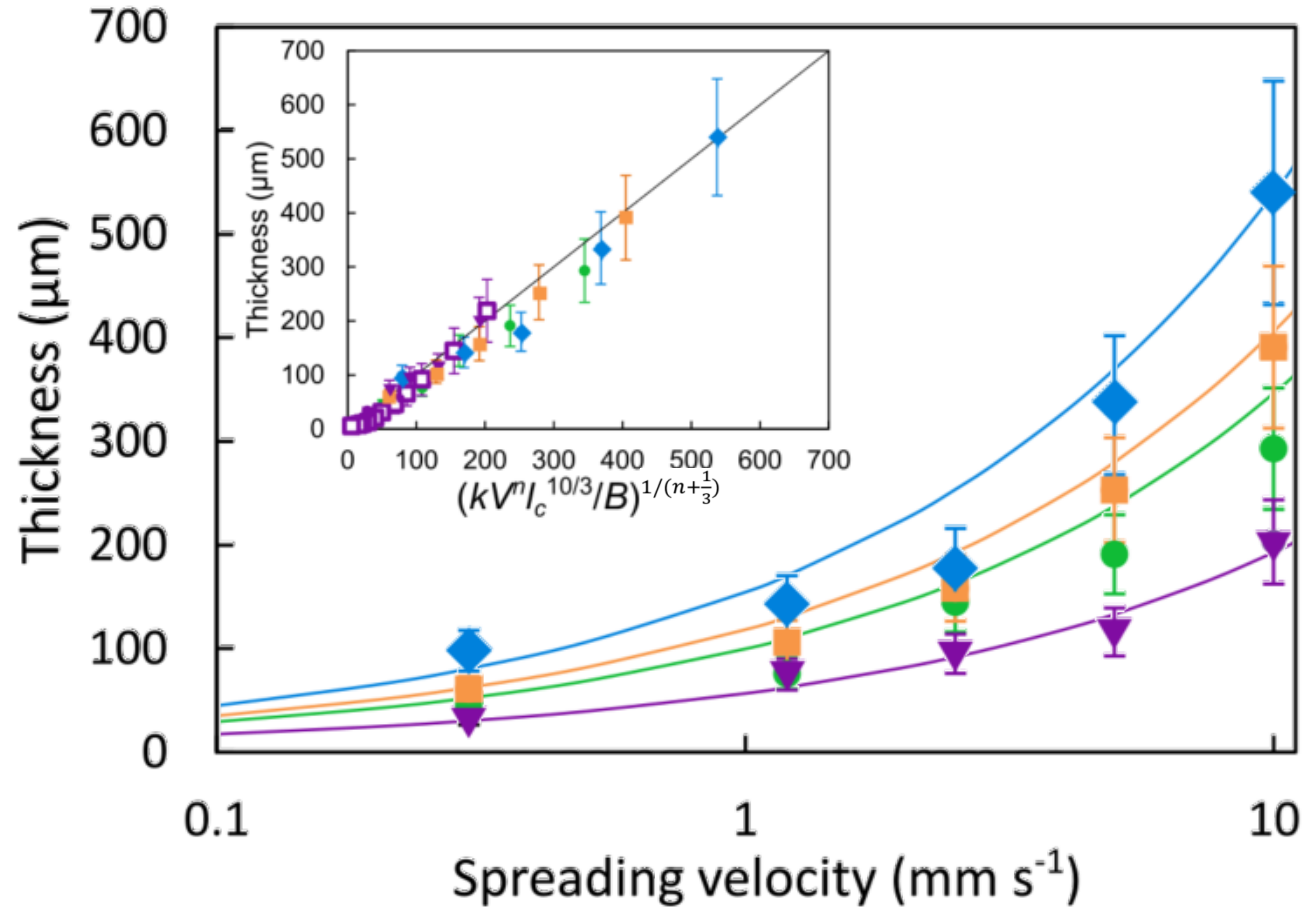
$B$  : Rigidity of the blade

$k$  : Viscosity parameter

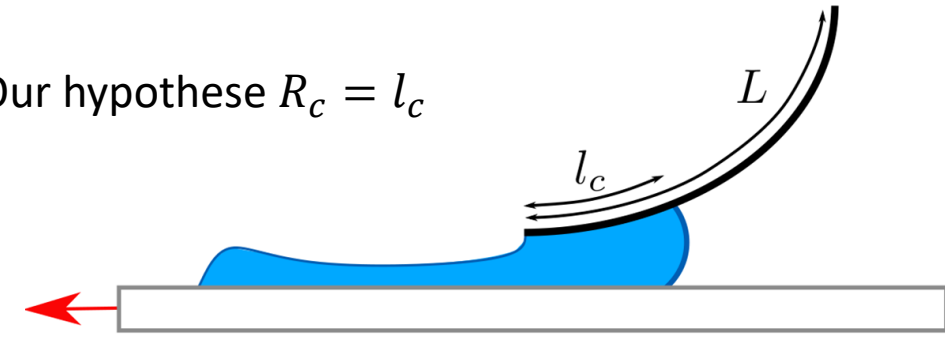
$V$  : Spreading velocity

# Blade Coating of carbopol gels

*Prediction of the film thickness*



Our hypothesis  $R_c = l_c$



- ▼ Carbopol 20 Pa 0.5 g
- Carbopol 20 Pa 1 g
- Carbopol 37 Pa 0.5 g
- ◆ Carbopol 47 Pa 0.5 g

$$h \propto \left( \frac{k V^n l_c^{10/3}}{B} \right)^{1/(n+\frac{1}{3})}$$

Modelling **consistent** with the experiments

# Blade Coating of carbopol gels

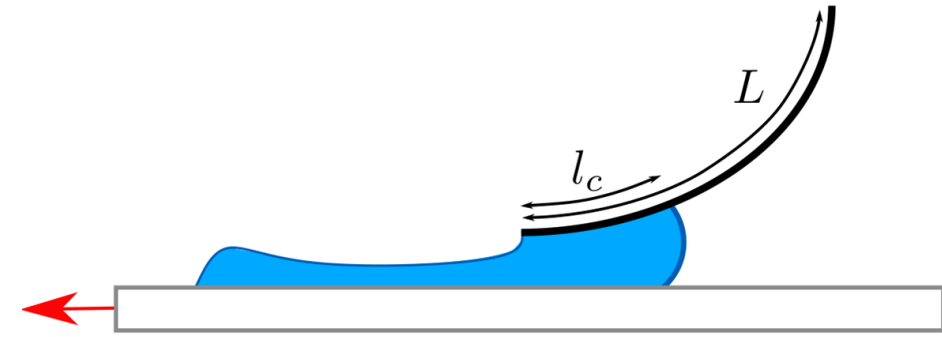
*Prediction of the film thickness*

$$h \propto \left( \frac{k V^n R_c^{10/3}}{B} \right)^{1/(n + \frac{1}{3})}$$

Newtonian case :  $n = 1$  (Seiwert)

$$e \sim L \left( \frac{\eta V L^2}{B} \right)^{3/4}$$

$R_c = L$  for Seiwert's model



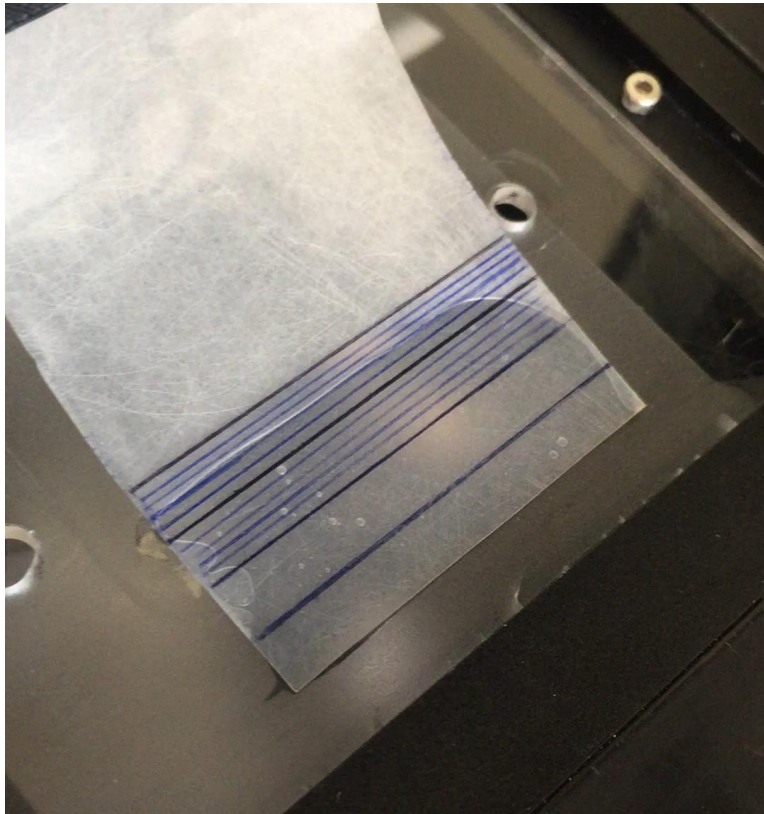
Not consistent with experiments

Cannot explain the **thickness evolution** during the spreading process

# Blade Coating of carbopol gels

*Prediction of the film thickness*

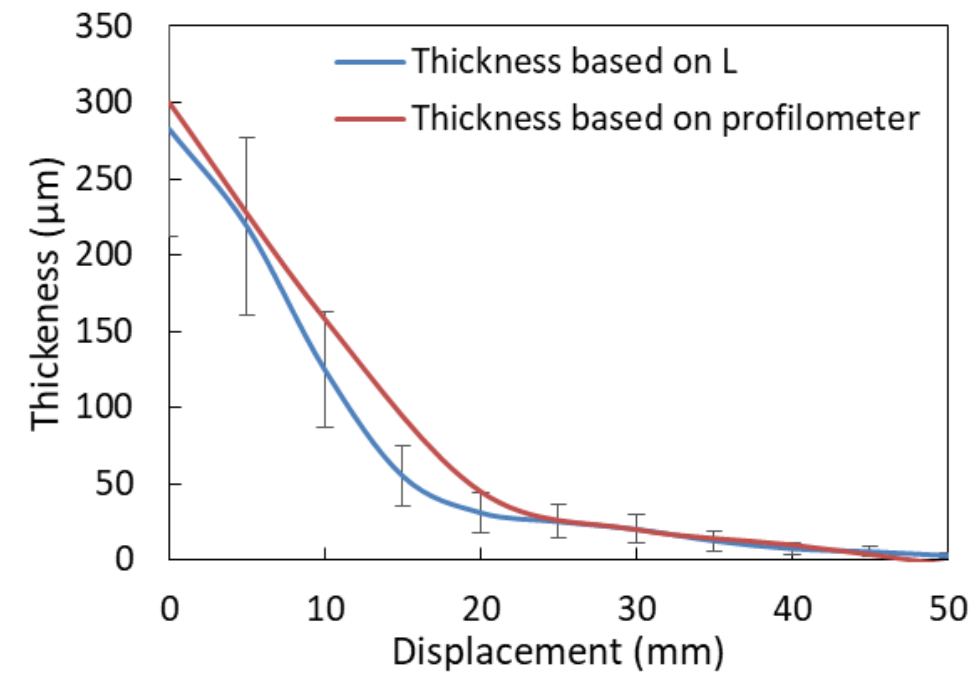
$$h \propto \left( \frac{k V^n l_c^{10/3}}{B} \right)^{1/(n+\frac{1}{3})}$$



Variable thickness due to the evolution of  $L$  during spreading



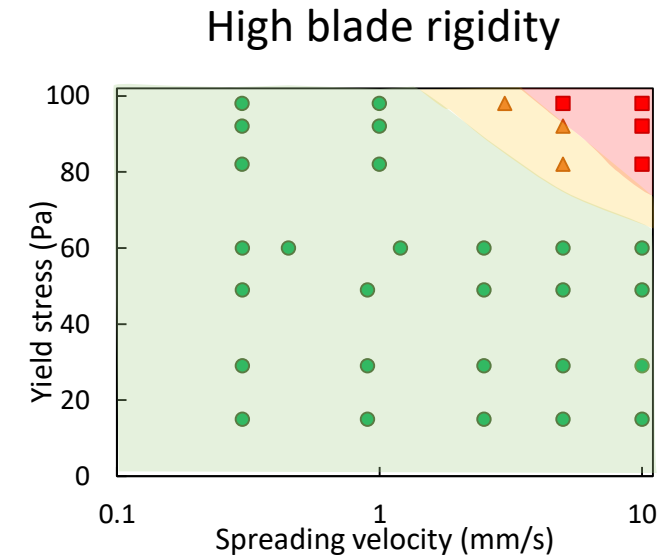
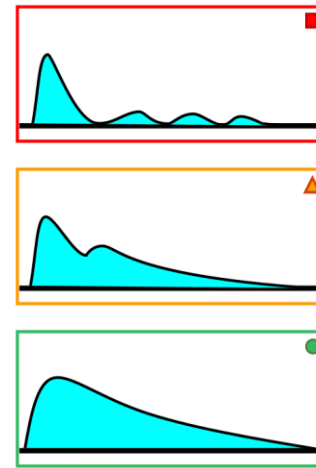
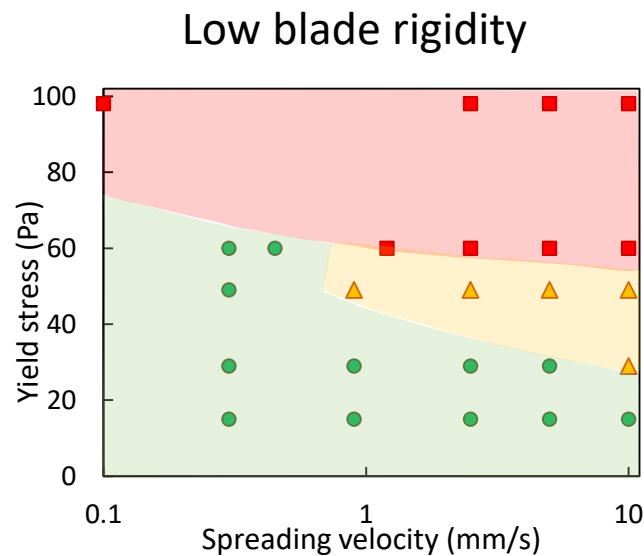
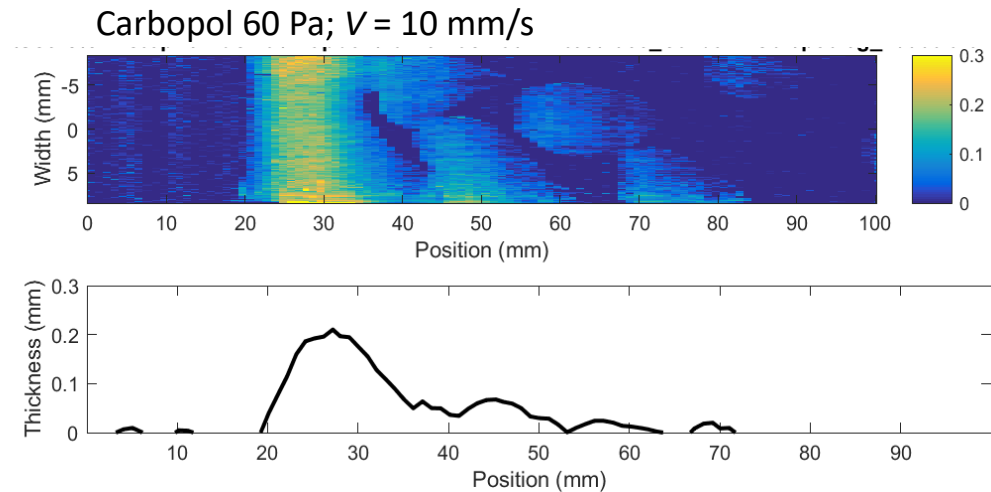
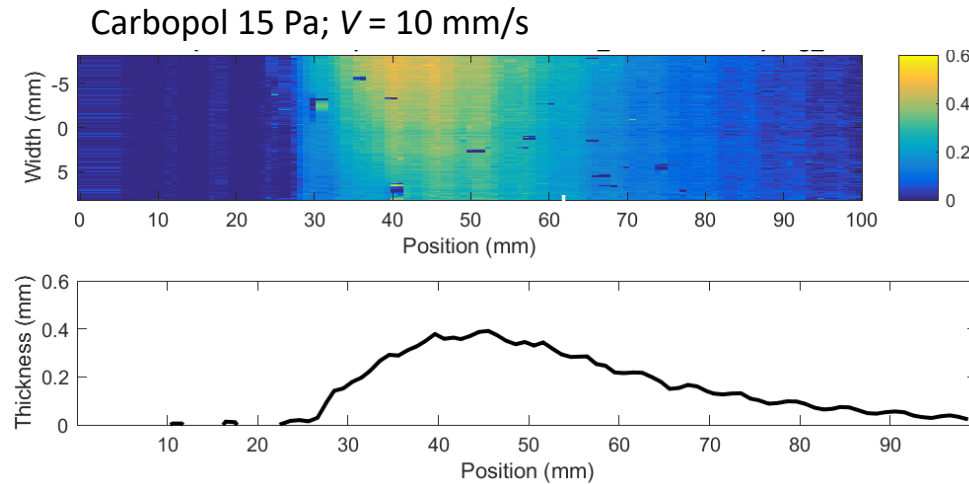
Decreasing of the thickness explained only if  $R_c = l_c$



Consistent with the use of  $R_c = l_c$

# Blade Coating of carbopol gels

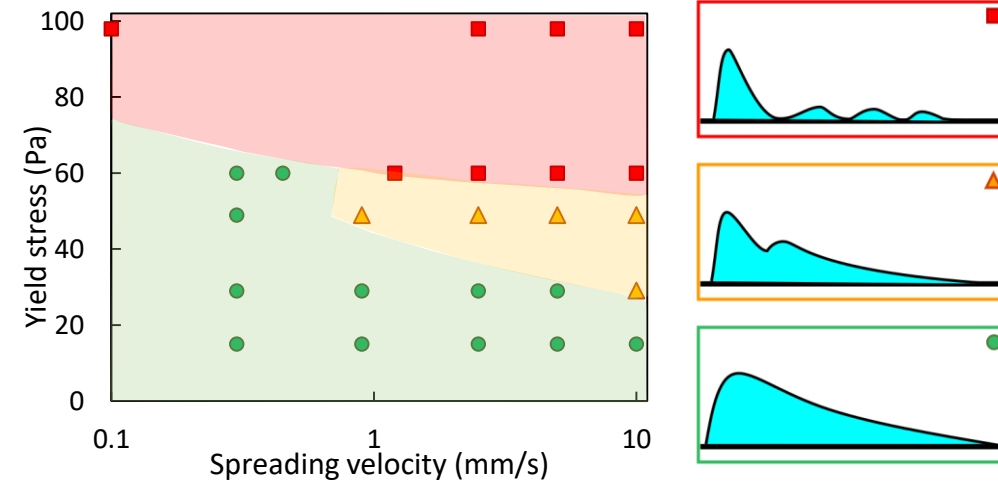
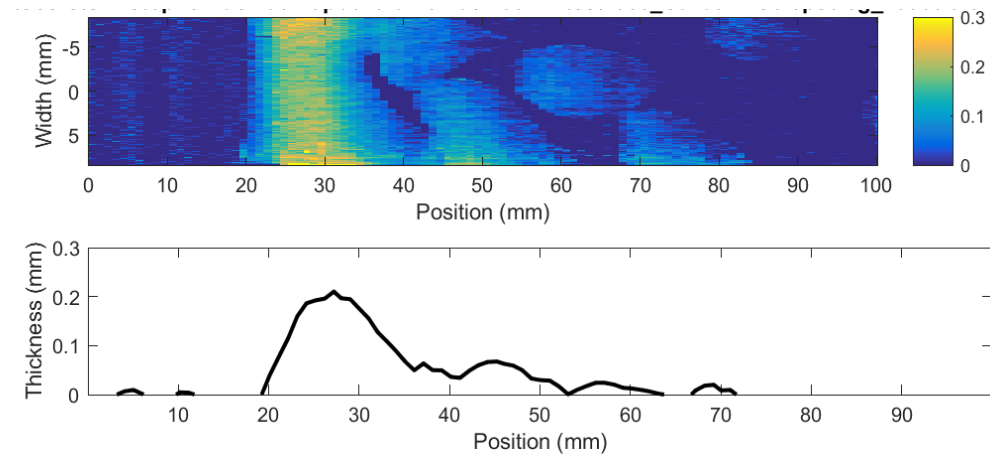
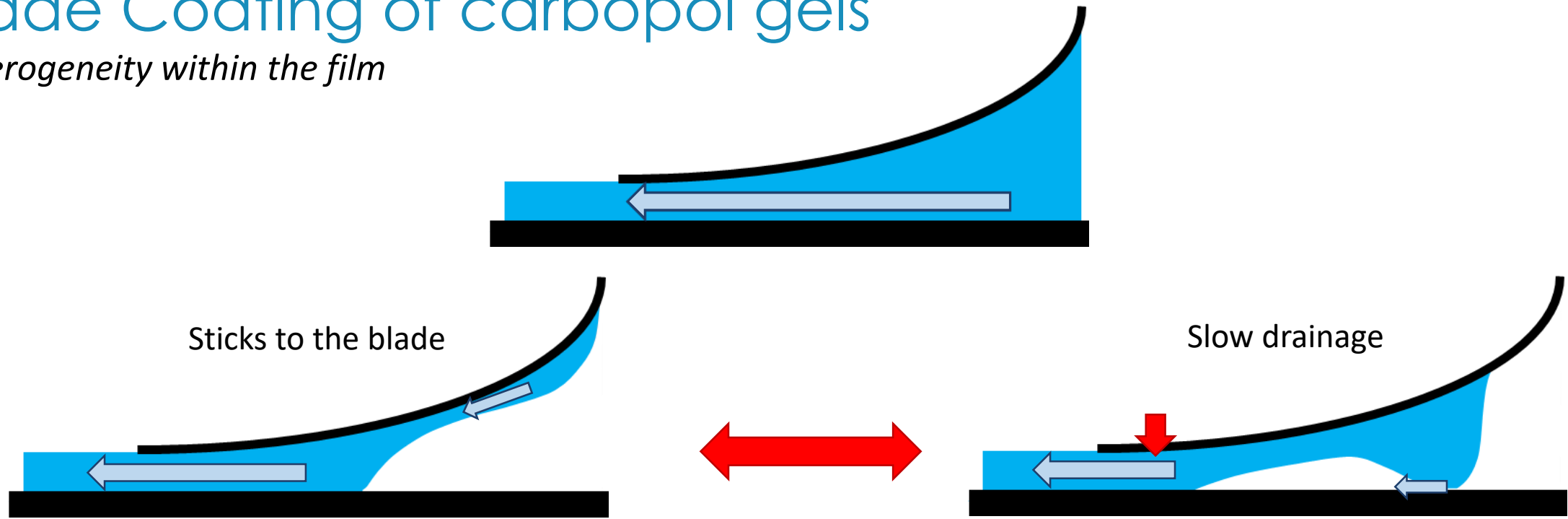
## *Heterogeneity within the film*



High spreading velocity and yield stress induce heterogeneity

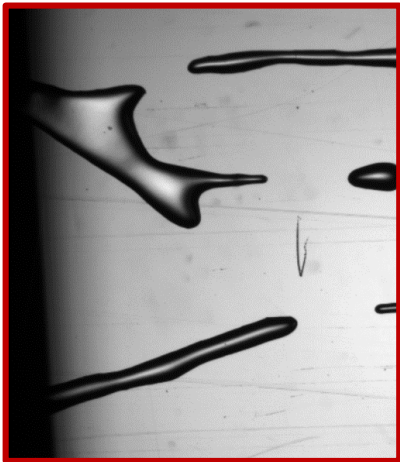
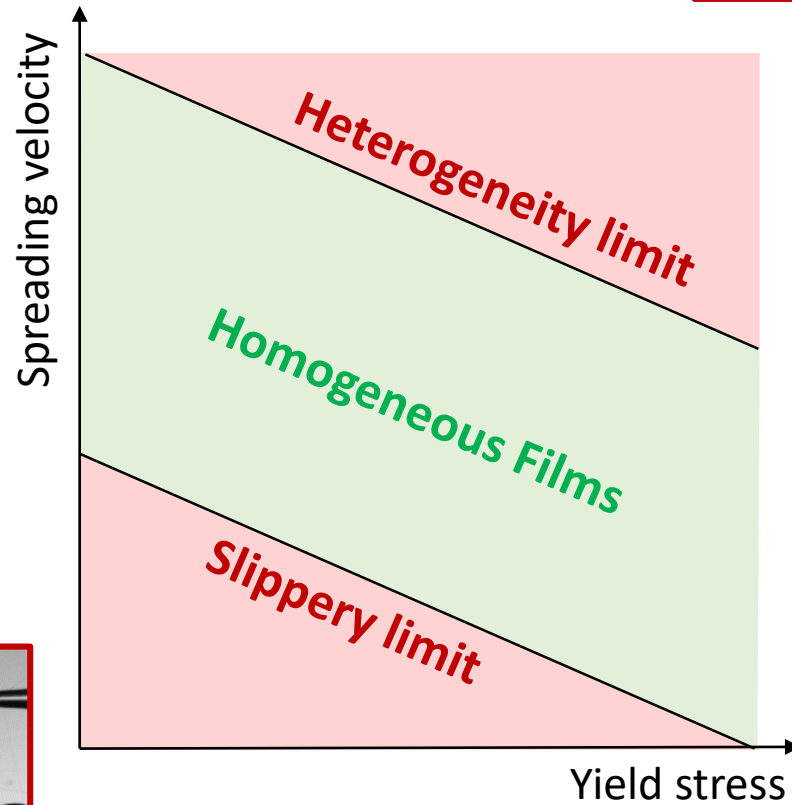
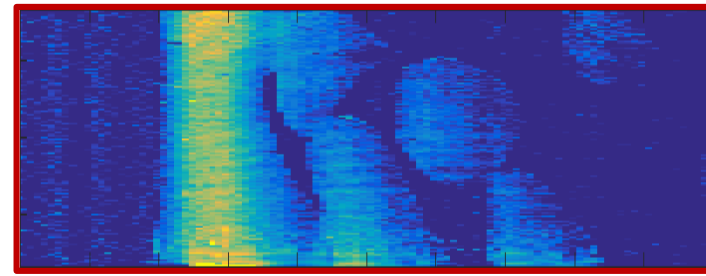
# Blade Coating of carbopol gels

*Heterogeneity within the film*



# Conclusion

The skin is neither smooth nor a perfect wetting system



Specific to a formulation on a substrate

Gives formulation limits

$$h \propto \left( \frac{k V^n l_c^{10/3}}{B} \right)^{1/(n+\frac{1}{3})}$$



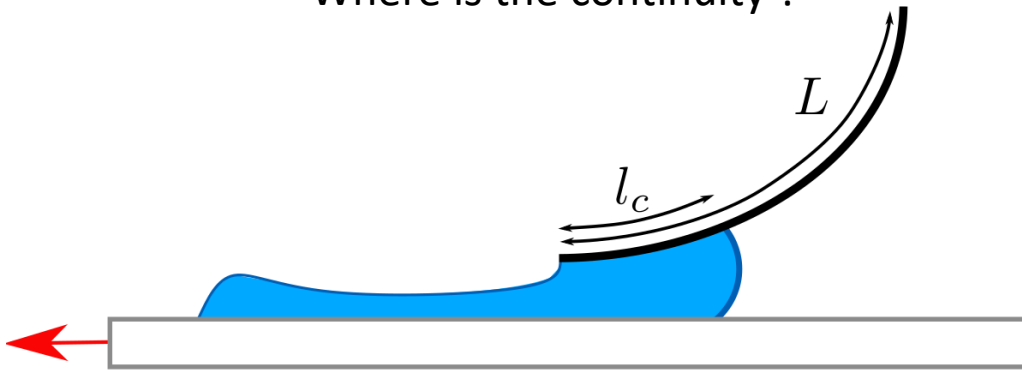
Able to **predict the film thickness** anytime during spreading using **Stokes' equation** and **geometrical parameters**

# Outlooks

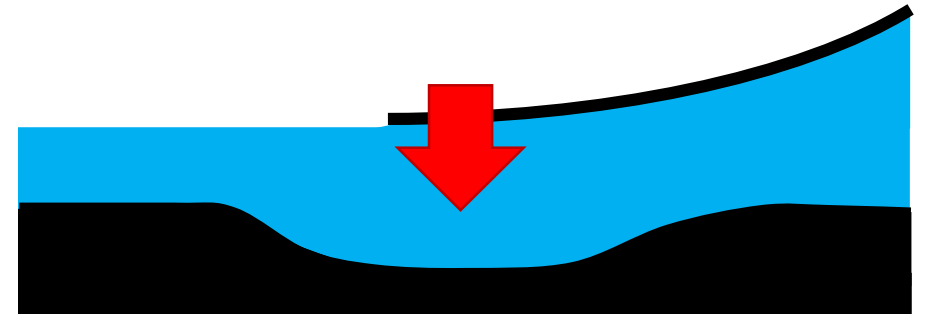
Seiwert  $R_c = L_{\text{blade}}$

Us  $R_c = l_c$

Where is the continuity ?



Spreading on soft substrate





**Part 2:**

***Complex formulations: spreading defects***

# Preparation of a model cosmetic formulation



Volatil phase (aqueous)



Water

Active non-volatil phase

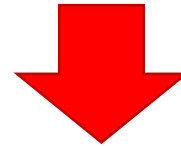


Glycerol, oils, waxes  
Pigments, UV filters

Texture agents



Carbopol  
Particles (starch, silica)



|                |                    |
|----------------|--------------------|
| Carbopol       | 0.3 % <sub>w</sub> |
| Water/Glycerol | 50-50 or 90-10     |
| Starch         | 6-13% <sub>v</sub> |

Hundreds of ingredients

# Impact of glycerol and cornstarch

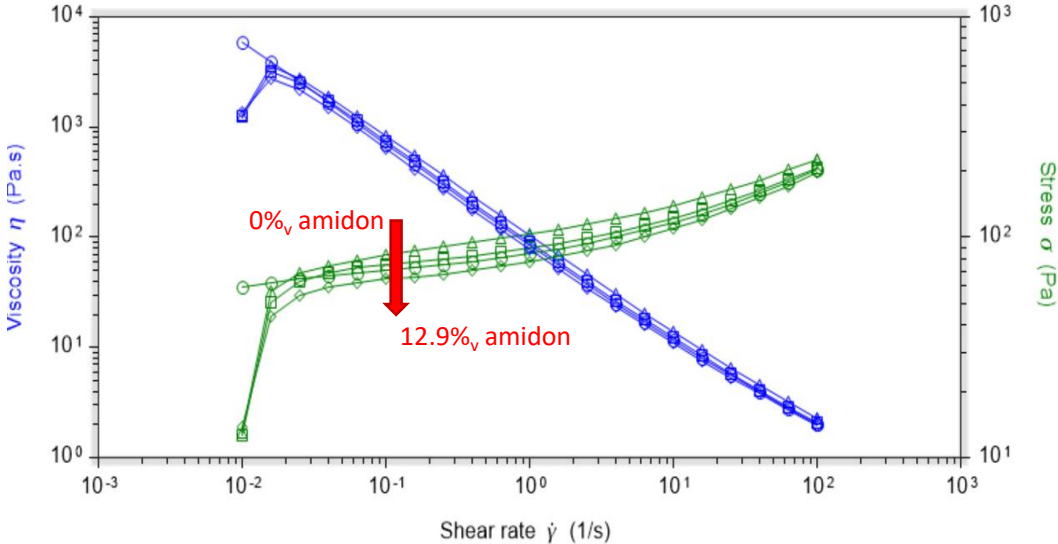
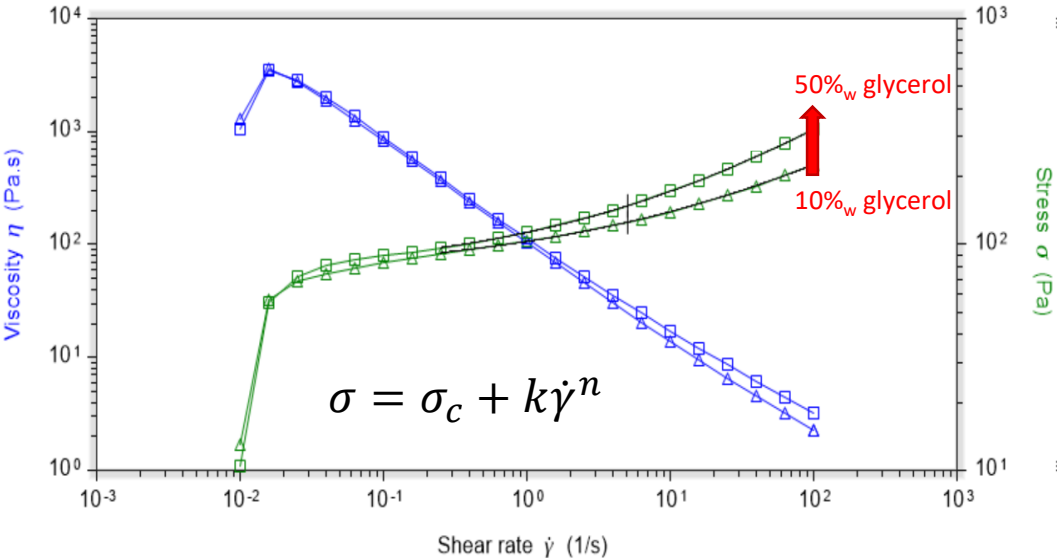
| Formula | % <sub>w</sub> carbopol<br>(in liquid phase) | % <sub>v</sub> starch<br>(global volume) | % <sub>w</sub> glycerol<br>(in liquid phase) | $\tau_y$<br>(Pa) | $k$ (Pa.s <sup>1/n</sup> ) | $n$  | $\varphi_{starch}$<br>(in the dry film) |
|---------|----------------------------------------------|------------------------------------------|----------------------------------------------|------------------|----------------------------|------|-----------------------------------------|
| 1       |                                              | 6.4                                      | 10                                           | 50.8             | 18.4                       | 0.40 | 0.46                                    |
| 2       |                                              | 9.3                                      |                                              | 48.3             | 16.9                       | 0.42 | 0.56                                    |
| 3       | 0.3                                          | 12.0                                     |                                              | 44.0             | 16.0                       | 0.44 | 0.63                                    |
| 4       |                                              | 6.9                                      |                                              | 46.8             | 29.1                       | 0.44 | 0.14                                    |
| 5       |                                              | 10.0                                     | 50                                           | 41.5             | 27.4                       | 0.45 | 0.20                                    |
| 6       |                                              | 12.9                                     |                                              | 37.6             | 25.7                       | 0.47 | 0.25                                    |



Starch particles unstable in water  
Experiments carried within 1 week



Highest and lowest  
volume fraction



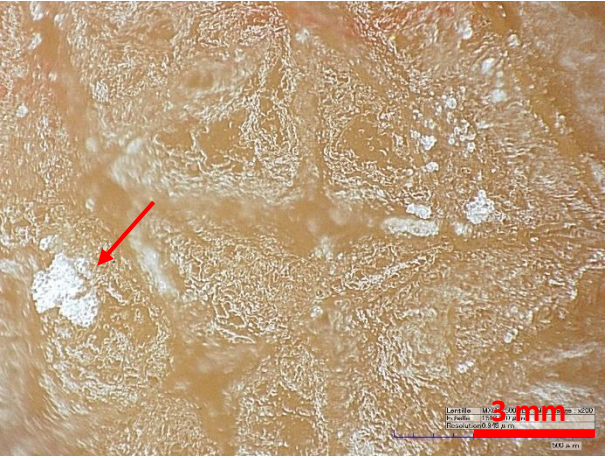
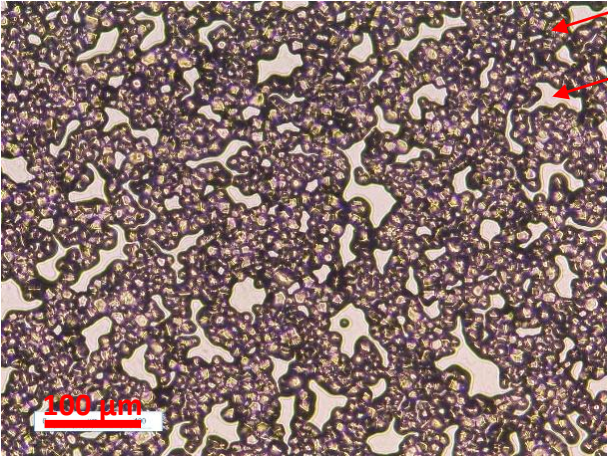
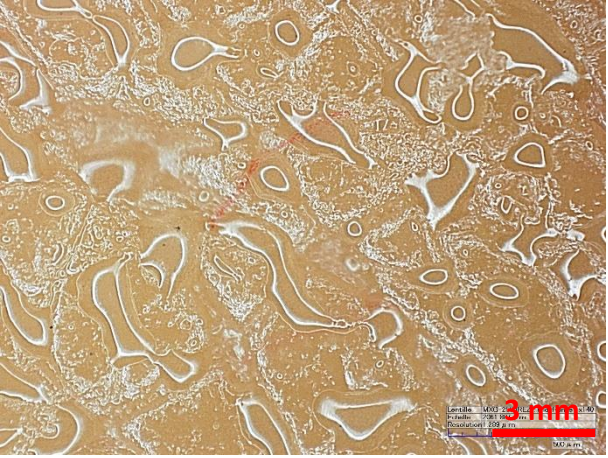
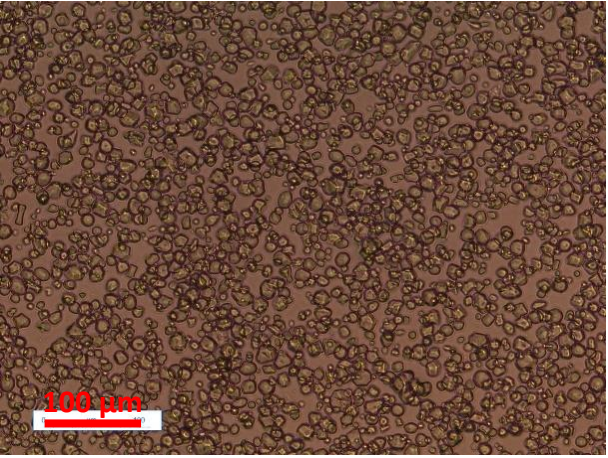
The glycerol increases the viscosity of the media

The starch decreases the yield stress of the media

# Impact of glycerol and cornstarch

## Structure of a dry deposit

| Formula | % <sub>w</sub> carbopol<br>(in liquid phase) | % <sub>v</sub> starch<br>(global volume) | % <sub>w</sub> glycerol<br>(in liquid phase) | $\phi_{\text{starch}}$<br>(in the dry film) | $\tau_y$<br>(Pa) | $k$ (Pa.s <sup>1/n</sup> ) | $n$  |
|---------|----------------------------------------------|------------------------------------------|----------------------------------------------|---------------------------------------------|------------------|----------------------------|------|
| 1       | 0.3                                          | 6.4                                      | 10                                           | 0.46                                        | 50.8             | 18.4                       | 0.40 |
| 2       |                                              | 9.3                                      |                                              | 0.56                                        | 48.3             | 16.9                       | 0.42 |
| 3       |                                              | 12.0                                     |                                              | 0.63                                        | 44.0             | 16.0                       | 0.44 |
| 4       | 0.3                                          | 6.9                                      | 50                                           | 0.14                                        | 46.8             | 29.1                       | 0.44 |
| 5       |                                              | 10.0                                     |                                              | 0.20                                        | 41.5             | 27.4                       | 0.45 |
| 6       |                                              | 12.9                                     |                                              | 0.25                                        | 37.6             | 25.7                       | 0.47 |

|                                             | Spreading on artificial skin                                                       | Single-layer coating on a glass plate                                               |
|---------------------------------------------|------------------------------------------------------------------------------------|-------------------------------------------------------------------------------------|
| $\phi_{\text{starch}} \text{ (dry)} = 0.63$ |   |   |
| $\phi_{\text{starch}} \text{ (dry)} = 0.14$ |  |  |

Particles

Air

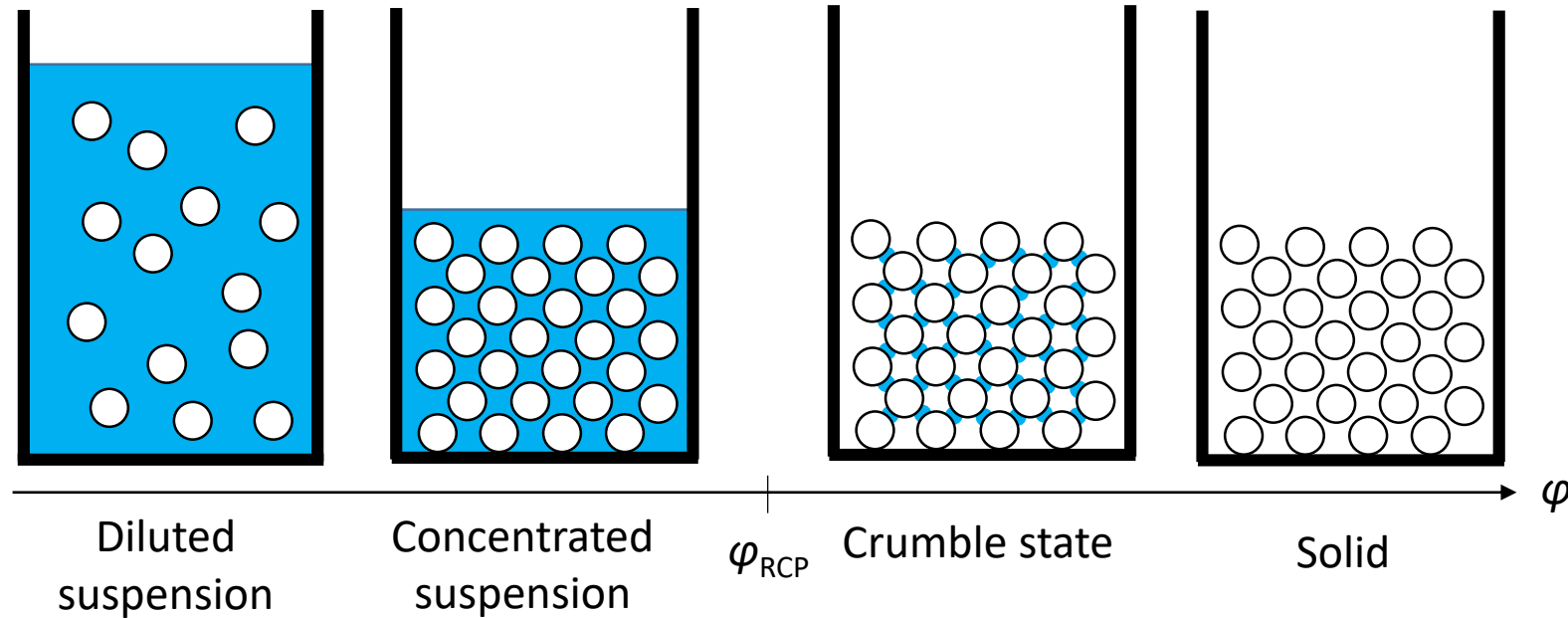
Capillary bridges formed by glycerol  
**Formation of Aggregates !**

Aggregates are formed at **high volume fraction** of particles

Starch particles **still dispersed** in the remaining glycerol

# The volume fraction in the dry film, a key parameter

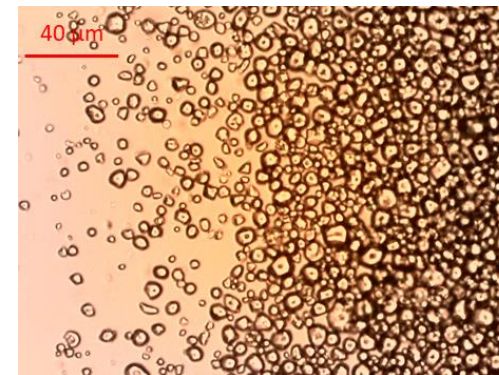
What does a formula look like as a function of the particle volume fraction?



$\phi_{\text{Close packing (perfect spheres)}} = 0.74$

$\phi_{\text{Random close packing (perfect spheres)}} = 0.64$

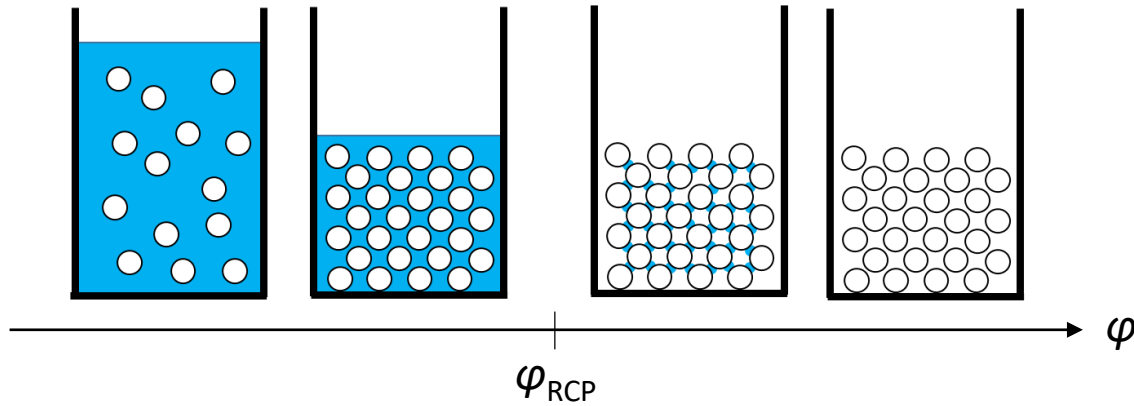
- Shape of particules
  - Polydispersity
- 
- Nature of the solvent
  - Friction coefficient of particles



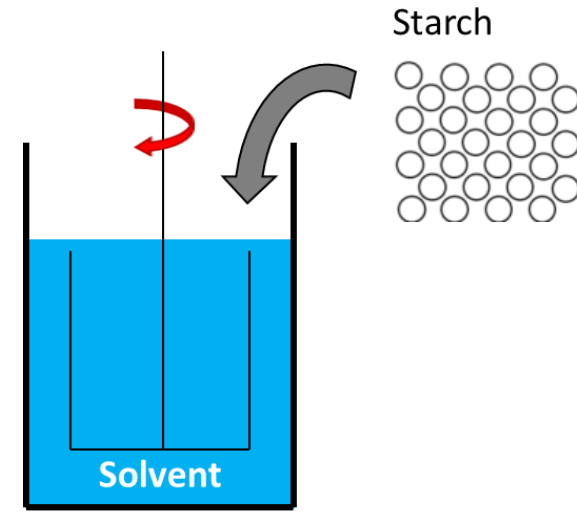
Has to be determined more precisely

# The volume fraction in the dry film, a key parameter

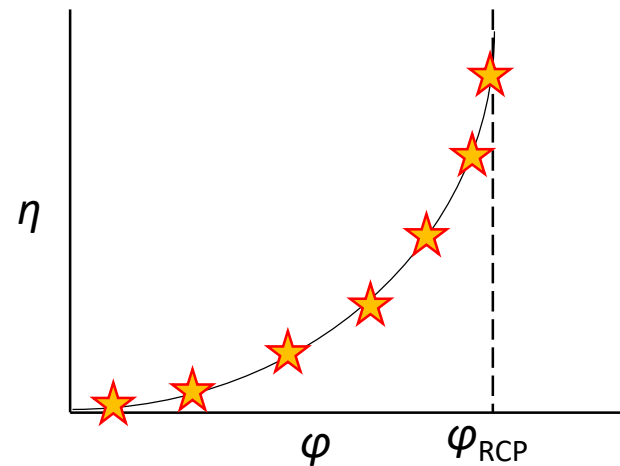
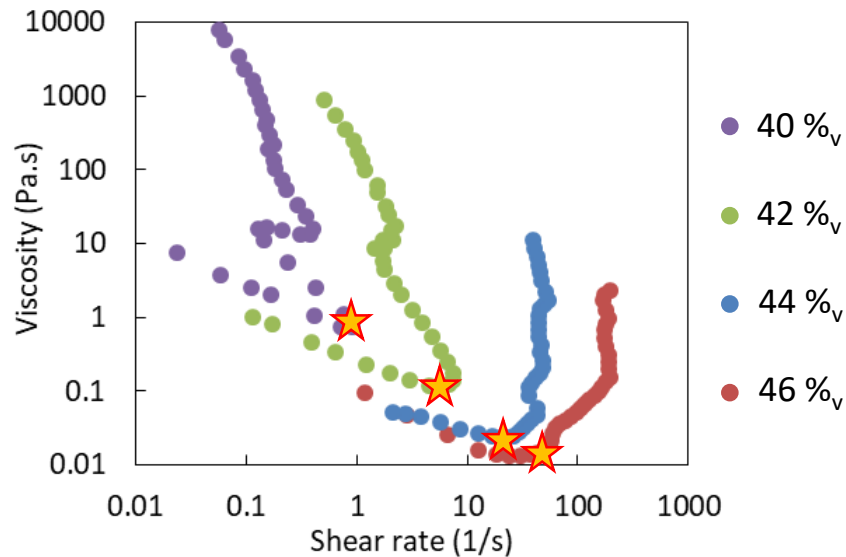
How do we measure  $\varphi_{RCP}$  ? Two methods



$\varphi_{RCP}$  (water) = 50 %<sub>v</sub>    $\varphi_{RCP}$  (glycerol) = 56 %<sub>v</sub>



Important tool in formulation science



Krieger-Dougherty model

$$\eta = \eta_0 \left( 1 - \frac{\varphi}{\varphi_{RCP}} \right)^{-[\eta]\varphi_{RCP}}$$

$[\eta]$  : Intrinsic viscosity

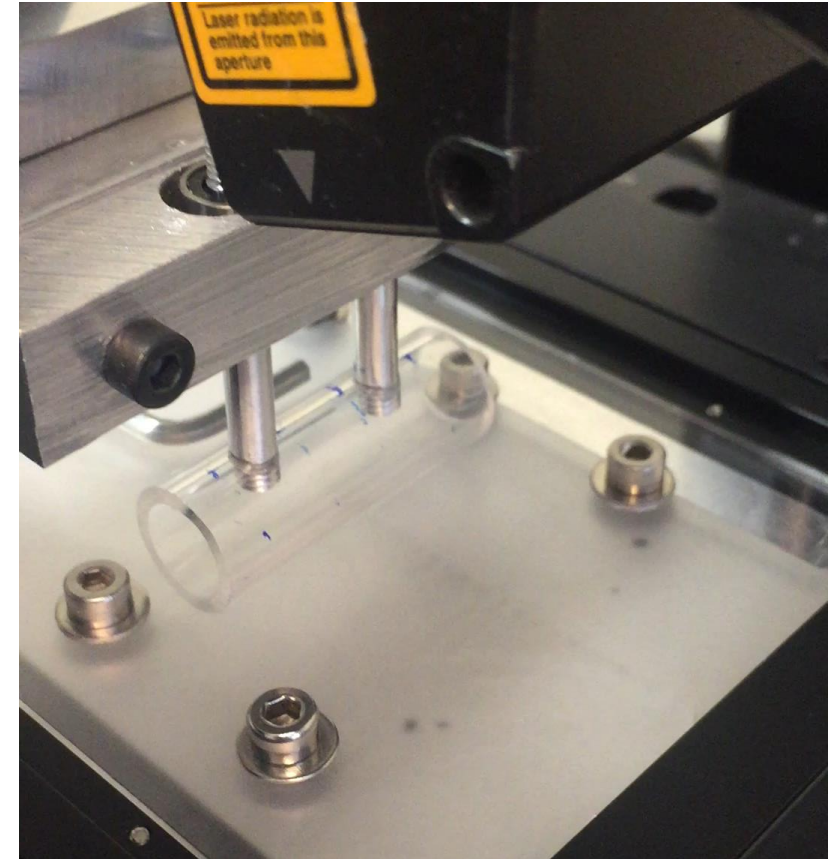
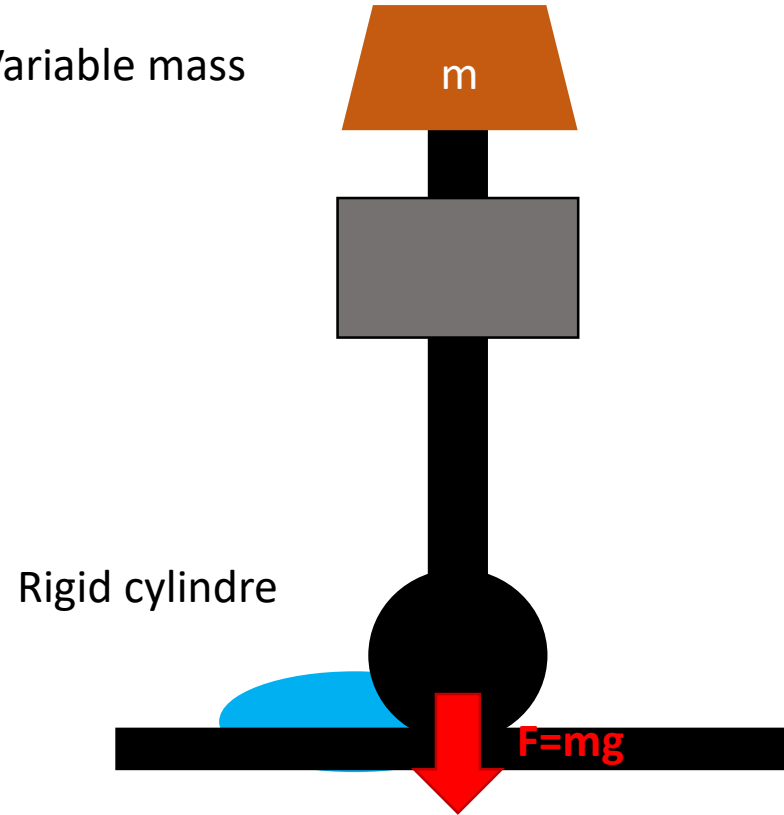
$\eta_0$  : Viscosity of the solvent

# Carrying more realistic spreading experiments

*Evolution of the experimental set-up*



Variable mass

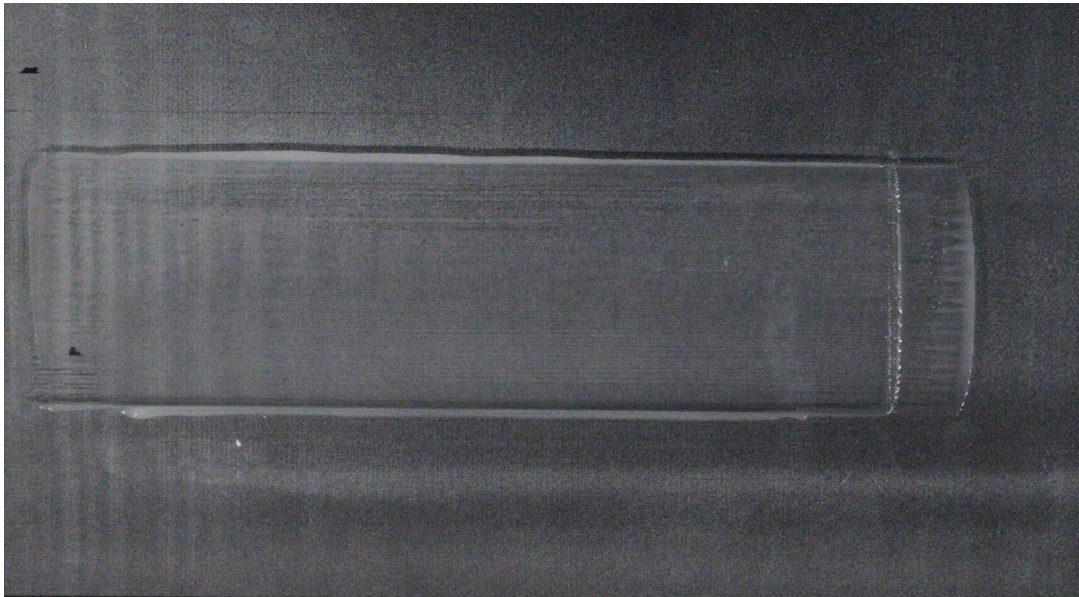


# Carrying more realistic spreading experiments

## *New spreading experiment*

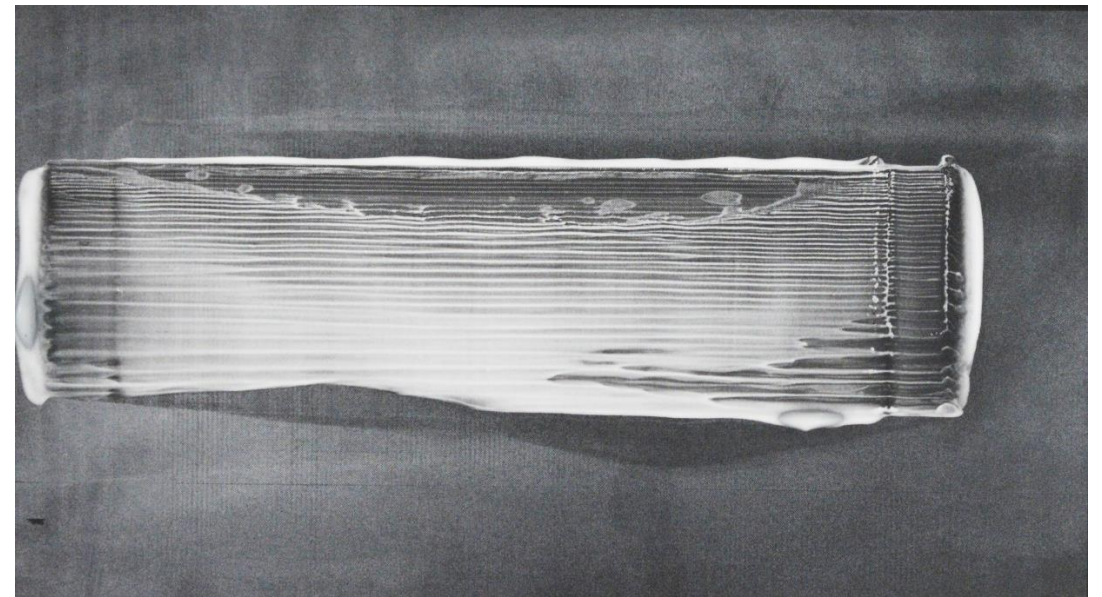
m sample = 0,5g  
Application force = 63mN →  $\left. \begin{array}{l} 15 \text{ Back and forth} \\ V = 10 \text{ mm/s} \end{array} \right\} \rightarrow 5 \text{ min drying} \right) \times 2$

$\varphi_{starch} \text{ (dry)} = 0.14$



Fluid and homogeneous coating

$\varphi_{starch} \text{ (dry)} = 0.63$



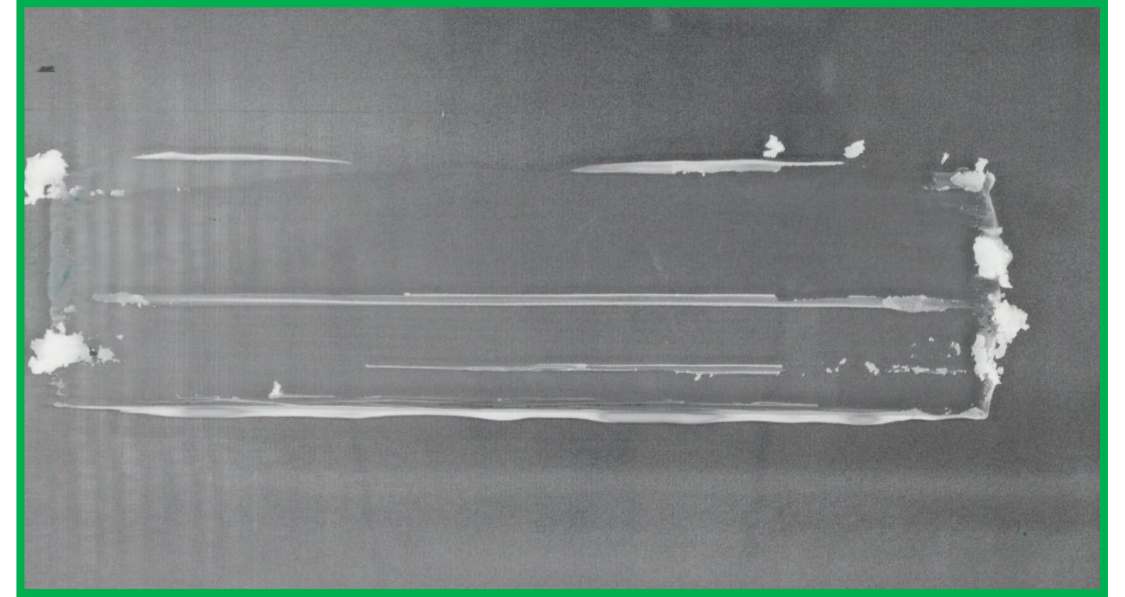
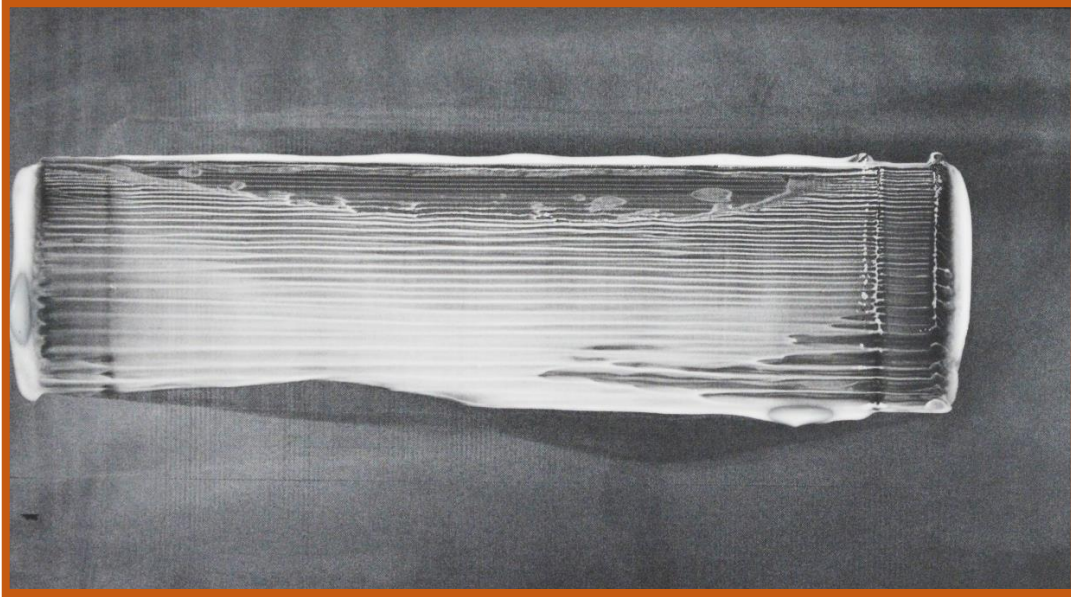
Dry and heterogeneous coating

Impact of  $\varphi$  highlighted

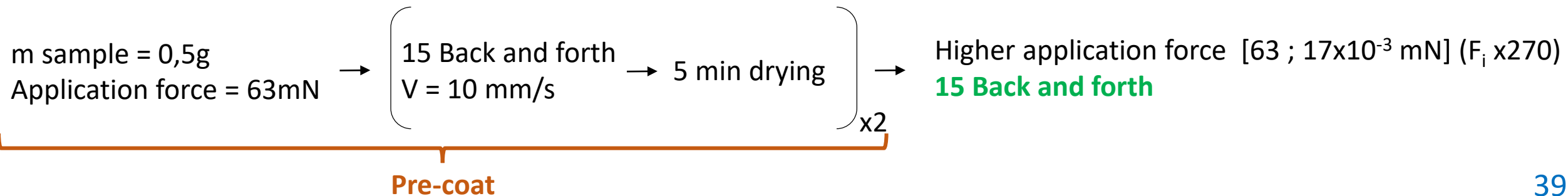
# Carrying more realistic spreading experiments

## New spreading experiment

$$\varphi_{starch}(\text{dry}) = 0.63$$



Coating peeled off through successive back and forth movements



# Carrying more realistic spreading experiments

*Three materials*



Smooth PMMA



Rough PMMA



Artificial leather

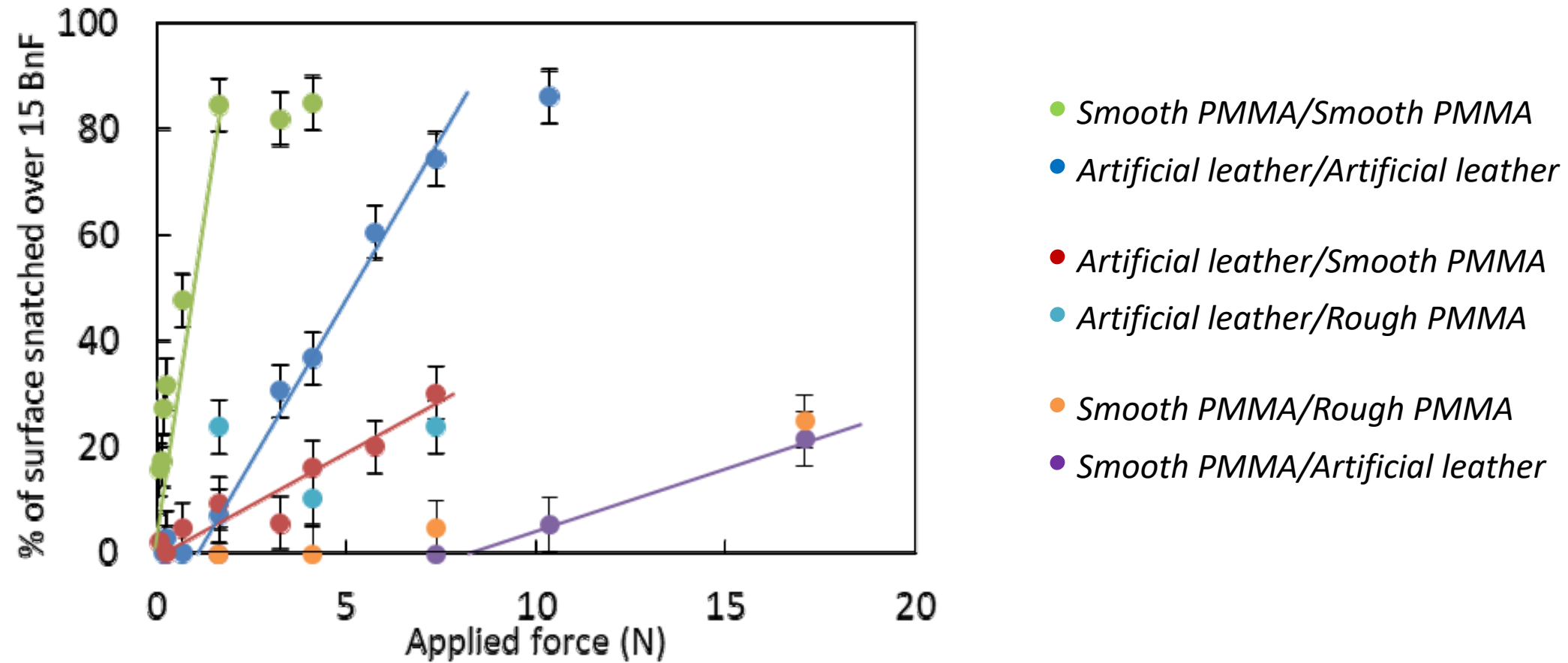


Checking the impact of  
roughness and softness

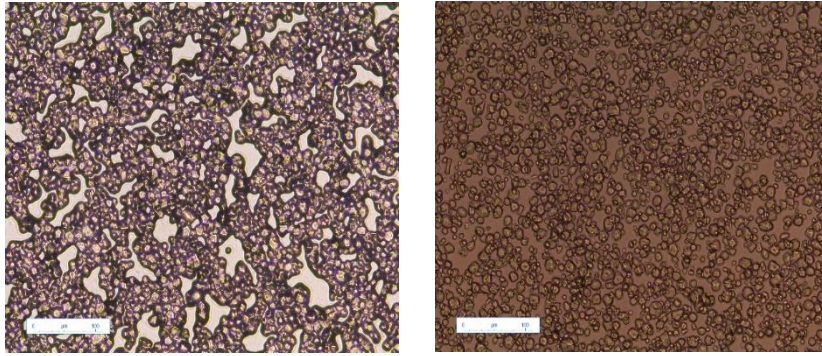


# Carrying more realistic spreading experiments

*Impact of the materials on the spreading defects*



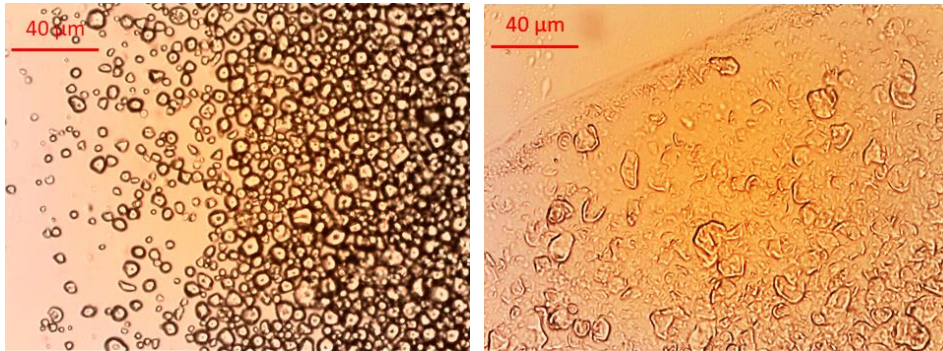
# Conclusion and outlooks



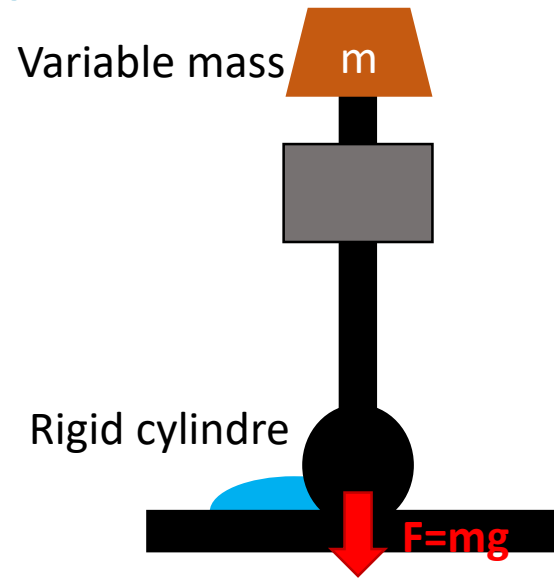
Impact of  $\varphi$   
Critical parameter  $\varphi_{RCP}$



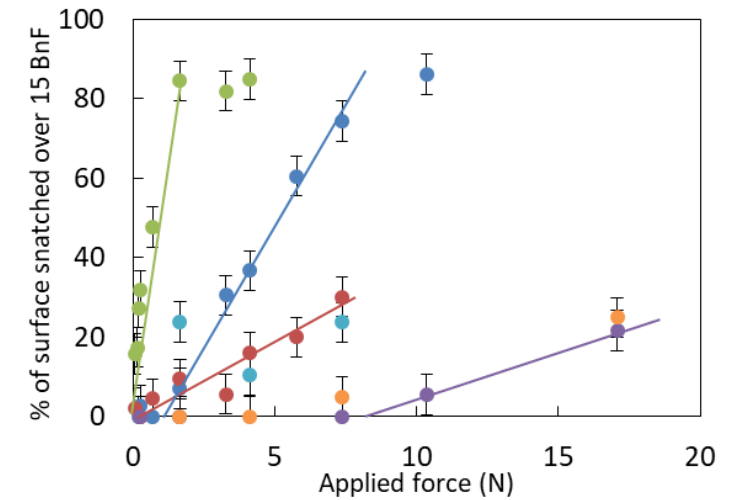
Evolution of  $\varphi$  overtime



Starch before and after ageing in water



More realistic  
spreading set-up



Impact of the substrate and  
applicator on spreading defects

Upgrade experimental set-up

- xyz force captor (stress)
- use even more realistic materials
- adapt the gesture

# *Conclusion et perspectives*

How are films formed using an elastic blade?

How can we predict film thickness for complex fluids?

What do I get if I spread very complex fluids?

# ACKNOWLEDGMENTS

JURY MEMBERS: CATHERINE BARENTIN,  
GUILLAUME OVARLEZ, JOSÉ BICO, DANIEL BONN

L'ORÉAL: ODILE AUBRUN, JEAN-BAPTISTE BOITTE,  
DANIELLE LE VERGE

CBI LABORATORY: JÉRÔME BIBETTE, ISABELLE BORSENBERGER, HÉLÈNE DODIER

MIE TEAM: ANNIE COLIN, WILBERT SMIT, ...

WORKSHOP TEAM: ALEXANDRE LANTHEAUME, LUDOVIC OLANIER, AMAURY FOURGEAUD

